

Honors Chemistry  
Practice Test #4

Name \_\_\_\_\_  
Date \_\_\_\_\_

Multiple Choice

Identify the choice that best completes the statement or answers the question.

- Which one of the following compounds is not a salt?
  - $\text{Ag}_2\text{SO}_4$
  - $\text{Ca(OH)}_2$
  - $\text{KCl}$
  - $\text{Ca(NO}_3)_2$
  - $\text{NaC}_2\text{H}_3\text{O}_2$
- Which of the following is the BEST definition of a salt?
  - A substance that produces ions in solution.
  - A substance that is completely ionized in solution.
  - A compound composed of metals and nonmetals.
  - A compound that contains a cation other than  $\text{H}^+$  and an anion other than  $\text{OH}^-$ .
  - A substance that does not produce ions in solution.
- Which one of the following compounds is a salt?
  - $\text{HClO}_2$
  - $\text{HC}_2\text{H}_3\text{O}_2$
  - $\text{HCN}$
  - $\text{NH}_4\text{C}_2\text{H}_3\text{O}_2$
  - $\text{KOH}$
- Which one of the following is a strong acid?
  - $\text{HF}$
  - $\text{HNO}_3$
  - $\text{HC}_2\text{H}_3\text{O}_2$
  - $\text{H}_2\text{SO}_3$
  - $\text{H}_2\text{CO}_3$
- Which statement regarding nitric acid is false?
  - It only slightly ionizes in solution.
  - Its solutions conduct electricity.
  - It is soluble in water.
  - It is a strong electrolyte.
  - It produces  $\text{H}^+$  and  $\text{NO}_3^-$  in solution.
- Which one of the following is a weak acid?
  - $\text{HBr}$
  - $\text{HNO}_3$
  - $\text{HClO}_4$
  - $\text{HCl}$
  - $\text{HClO}_2$
- Which one of the following is a strong and soluble base?
  - $\text{Ni(OH)}_2$
  - $\text{Mg(OH)}_2$
  - $\text{BaOH}$
  - $\text{Ca(OH)}_2$
  - $\text{Fe(OH)}_3$

# Honor Chemistry Practice Test

**Hussin A.Rothana**



## **Honor Chemistry Practice Test:**

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