

HONORS CHEMISTRY INTRODUCTION TO RATES AND EQUILIBRIUM

PART 1: COLLISION THEORY

Follow the links below to complete the assignment. Some animations may take a while to load so be patient!

A. **Watch:** Ted ED: "How to get a Date"

<http://ted.led.com/essays/how-to-speed-up-chemical-reactions-and-get-a-date>

- While watching, take 10 bullet points of notes on how to speed up chemical reactions.
- Click "Think" tab and take the quiz. Write the answers here in complete sentences (yes, I mean it this time).

B. **Visit:** [The Collision Theory of Reaction Rates](http://www.chemguide.co.uk/physical/basicrates/introduction.html) -

<http://www.chemguide.co.uk/physical/basicrates/introduction.html>

- Read the information describing the collision theory. Answer the questions that follow.
1. It is pretty obvious that if you have a situation involving two reactants they can only react together if they come into contact with each other. They first have to collide, and then they **may** react. Why is the term "**may** react" used?
 2. What is activation energy?
 3. Describe the significance of the Maxwell - Boltzmann Distribution Curve.

C. **Watch:** Orientation of Collisions - see link below

<http://www.mhhe.com/physac/chemistry/essentialchemistry/flash/colls11.swf>

Explain, in your own words, what "Orientation of Collisions" means.

D. **Watch:** [Activation Energy](#)

<http://www.mhhe.com/physac/chemistry/essentialchemistry/flash/activa2.swf>

Sketch a potential energy diagram of the two following reaction coordinates (endothermic and exothermic). Label all parts and briefly describe what is happening at each part of the reaction coordinate curve (A, B, C, and D):

An Endothermic reaction



An Exothermic Reaction



Honors Chemistry Equilibrium

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Principles of Modern Chemistry David W. Oxtoby, H. Pat Gillis, Laurie J. Butler, 2016-01-01 Long considered the standard for honors and high level mainstream general chemistry courses PRINCIPLES OF MODERN CHEMISTRY continues to set the standard as the most modern rigorous and chemically and mathematically accurate text on the market This authoritative text features an atoms first approach and thoroughly revised chapters on Quantum Mechanics and Molecular Structure Chapter 6 Electrochemistry Chapter 17 and Molecular Spectroscopy and Photochemistry Chapter 20 In addition the text utilizes mathematically accurate and artistic atomic and molecular orbital art and is student friendly without compromising its rigor End of chapter study aids focus on only the most important key objectives equations and concepts making it easier for students to locate chapter content while applications to a wide range of disciplines such as biology chemical engineering biochemistry and medicine deepen students understanding of the relevance of chemistry beyond the classroom **Fast**

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