

(1) Given this equation: N2 + 3 H2 ---> 2 NH3, write the following molar ratios:

② Given the following equation: 8 H<sub>2</sub> + S<sub>8</sub> --> 8 H<sub>2</sub>S, write the following molar ratios:

(3) Answer the following questions for this equation: 2 H<sub>2</sub> + O<sub>2</sub> ---> 2 H<sub>2</sub>O

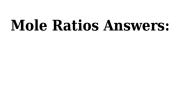
b) Suppose you had 20 moles of 
$$H_2$$
 on hand and plenty of  $O_2$ , how many moles of  $H_2O$  could you make? 20 moles  $H_2$  2  $H_2O$  = 20 moles  $H_2O$ 

4) Use this equation: N2 + 3 H2 --> 2 NH3, for the following problems

d) If 0.600 moles of NH3 were produced, how many moles of H2 are required?

# **Mole Ratios Answers**

SJ Ball



Whispering the Techniques of Language: An Mental Quest through Mole Ratios Answers

In a digitally-driven earth wherever displays reign great and instant interaction drowns out the subtleties of language, the profound secrets and mental nuances concealed within phrases frequently move unheard. However, situated within the pages of **Mole Ratios Answers** a fascinating fictional value sporting with raw feelings, lies an extraordinary quest waiting to be undertaken. Published by a skilled wordsmith, that wonderful opus invites visitors on an introspective trip, gently unraveling the veiled truths and profound influence resonating within the very cloth of each and every word. Within the mental depths of this touching review, we shall embark upon a sincere exploration of the book is key styles, dissect its charming writing design, and fail to the effective resonance it evokes deep within the recesses of readers hearts.

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