THE MOLE AND AVOGADRO'S NUMBER

Name KEY

One mole of a substance contains Avogadro's Number (6.02 \times 10 23) of molecules.

How many molecules are in the quantities below?

1. 2.0 moles	2.0 mde	6.02 × 1023 mikum	= 1.20 ×1024 moderales
2. 1.5 moles	1.5 mag	6.02×10=3 moleculos =	9.03×1035 motals
3. 0.75 mole	0.75 mbc	6.02 NO23 mdeals	= 4.52 ×103 moderales
4. 15 moles	15 mbg	6.02×1023 moterals.	= 9.03×1024 mdeals
5. 0.35 mole	0.35 2	6.02×1033 mdeale	= 2.11 × 1023 moderals

How many moles are in the number of molecules below?

1.	6.02 x 10 ²³	I made (By definition!)
2.	1.204 x 1C ²⁴	1.204 × 1024 miterile mde = 2.00 mdes
3.	1.5 x 10 ²⁰	1.5 × 1025 1 mde = 2.49×10-4 mde
4.	3.4 x 10 ²⁶	34x1026 Imde = 565 mdes
5.	7.5 x 10**	15 x10 mode 1.24 x 10 modes

Mole And Avogadros Number Answer Key

R Barnett

Mole And Avogadros Number Answer Key:

Unveiling the Magic of Words: A Report on "Mole And Avogadros Number Answer Key"

In some sort of defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their ability to kindle emotions, provoke contemplation, and ignite transformative change is really aweinspiring. Enter the realm of "Mole And Avogadros Number Answer Key," a mesmerizing literary masterpiece penned with a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve into the book is central themes, examine its distinctive writing style, and assess its profound effect on the souls of its readers.

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