

Mole Calculation Practice Worksheet

Answer the following questions:

- 1) How many moles are in 25 grams of water?

$$\frac{25 \text{ g H}_2\text{O}}{18.02 \text{ g H}_2\text{O}} \times \frac{1 \text{ mol H}_2\text{O}}{1} = 1.4 \text{ moles}$$

- 2) How many grams are in 4.5 moles of Li_2O ?

$$\frac{4.5 \text{ mol Li}_2\text{O}}{1 \text{ mol Li}_2\text{O}} \times \frac{29.88 \text{ g Li}_2\text{O}}{1} = 130 \text{ g}$$

- 3) How many molecules are in 23 moles of oxygen?

$$\frac{23 \text{ mol O}_2}{1 \text{ mol O}_2} \times \frac{6.02 \times 10^{23} \text{ molecules}}{1} = 1.4 \times 10^{25} \text{ molecules}$$

- 4) How many moles are in 3.4×10^{23} molecules of H_2SO_4 ?

$$\frac{3.4 \times 10^{23} \text{ molecules H}_2\text{SO}_4}{6.02 \times 10^{23} \text{ molecules H}_2\text{SO}_4} \times \frac{1 \text{ mol H}_2\text{SO}_4}{1} = 0.56 \text{ moles}$$

- 5) How many molecules are in 25 grams of NH_3 ?

$$\frac{25 \text{ g NH}_3}{17.03 \text{ g NH}_3} \times \frac{1 \text{ mol NH}_3}{1} \times \frac{6.02 \times 10^{23} \text{ molecules NH}_3}{1 \text{ mol NH}_3} = 8.8 \times 10^{23} \text{ molecules}$$

- 6) How many grams are in 8.2×10^{22} molecules of N_2I_4 ?

$$\frac{8.2 \times 10^{22} \text{ molecules N}_2\text{I}_4}{6.02 \times 10^{23} \text{ molecules N}_2\text{I}_4} \times \frac{1 \text{ mol N}_2\text{I}_4}{1} \times \frac{789.42 \text{ g N}_2\text{I}_4}{1 \text{ mol N}_2\text{I}_4} = 110 \text{ g}$$

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Prof. Dr. ZİYA ENGİN ERKMEN



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Understanding Chemistry James Dudley Herron, 1981 CERAMIC MATERIALS AND PROCESSING - PHASE EQUILIBRIA - PHYSICAL MODELLING IN ELECTRON MICROSCOPY Prof. Dr. ZİYA ENGİN ERKMEN, 2019-09-15 There were two main driving forces in my decision for preparing a question answer book covering all the courses given by myself during the past 10 years in my academic career first argument is that there exists a good amount of original questions in the exams and their corresponding answers next my expectation is that probably fewer time would be spent in preparing such a

book where the questions and answers are all ready prepared thereupon In this country most of the undergraduate students do not necessarily work on the course they attend in the same day instead they prefer to start preparing their exams 2 or 3 days before In these circumstances last minute students may usually prefer working on the passed questions for the tomorrow s exam However this method of preparation may lead unwanted consequences such that students mostly do not find same questions unconsciously learnt by heart in the exams In order to increase working efficiency and consequently reach to the maximum performance in exams will only be possible if students do not miss any lectures given by staff in addition to be an active participant during the lecture and daily work on the given homeproblems are the basic requirements for a full success In concluding this book will give a chance to the well prepared students to make a quick rehearsal before the exams for obtaining best results after the exams

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