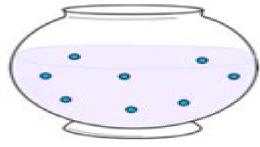
Introduction to Molarity

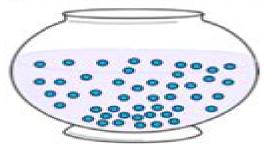
<u>Concentration</u> describes the amount of solute that is present in a certain amount of solution.

- > High concentration indicates a large amount of solute.
- Low concentration indicates a small amount of solute.

Example: Label the solutions as high concentration or low concentration.



Low Concentration



High Concentration

Molarity is used to measure the concentration of a solution.

Example: In every 5 liters of ocean water, there are 20 moles of NaCl (salt). What is the **molarity** of NaCl in sea water?

Molarity =
$$4.0 \text{ M} \text{ or mol/L}$$

Molarity By Dilution Worksheet Key

Karin Nielsen-Saines

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