Name: Date:

# **Limiting Reagent Worksheet**

Answers

Given the following reaction:

$$C_1H_A + \underline{5} O_1 \rightarrow \underline{3} CO_1 + \underline{4} H_1O_2$$

Balance the reaction.

Suppose 14.8 g of C<sub>3</sub>H<sub>8</sub> reacts with 3.44 g of O<sub>2</sub>. Determine the limiting reagent.

$$14.8 \text{ g C}_{3}H_{8} \times \frac{1 \text{ mol C}_{3}H_{8}}{44.09 \text{ g C}_{3}H_{8}} \times \frac{4 \text{ mol H}_{2}O}{1 \text{ mol C}_{3}H_{8}} \times \frac{18.02 \text{ g H}_{2}O}{1 \text{ mol H}_{2}O} = 24.2 \text{ g H}_{2}O$$

$$3.44 \text{ g O}_{2} \times \frac{1 \text{ mol O}_{2}}{32 \text{ g O}_{2}} \times \frac{4 \text{ mol H}_{2}O}{5 \text{ mol O}_{2}} \times \frac{18.02 \text{ g H}_{2}O}{1 \text{ mol H}_{2}O} = 1.55 \text{ g H}_{2}O$$

Since O, produces less quantity of H,O than C,H, does, it is the limiting reagent.

(b) Determine the amount of CO, produced.

Given the following reaction:

Balance the reaction.

What is the limiting reagent if 2.2 g of Mg reacts with 4.5 L of O, at STP?

2.2 g Mg \* 
$$\frac{1 \text{ mol Mg}}{24.3 \text{ g Mg}}$$
 \*  $\frac{2 \text{ mol MgO}}{2 \text{ mol Mg}}$  = 0.0905 mol MgO  
4.5 L O<sub>2</sub> \*  $\frac{1 \text{ mol O}_2}{22.4 \text{ L O}_2}$  \*  $\frac{2 \text{ mol MgO}}{1 \text{ mol O}_2}$  \* = 0.401 mol MgO

Since Mg produces less MgO than O, Mg is the limiting reagent.

Determine the amount of MgO produced in grams.

# **Limiting Reactant Work Sheet**

Arthur E. Morris, H. Alan Fine, Gordon Harold Geiger



Unveiling the Magic of Words: A Report on "Limiting Reactant Work Sheet"

In a global defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their power to kindle emotions, provoke contemplation, and ignite transformative change is really awe-inspiring. Enter the realm of "**Limiting Reactant Work Sheet**," a mesmerizing literary masterpiece penned by a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve to the book is central themes, examine its distinctive writing style, and assess its profound affect the souls of its readers.

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