

Name _____ Date _____ Class _____

CHAPTER 15 STUDY GUIDE FOR CONTENT MASTERY

Solutions

Section 15.1 What are solutions?

In your textbook, read about the characteristics of solutions.

Use each of the terms below just once to complete the passage.

immiscible	liquid	soluble	solution
insoluble	miscible	solvent	solvent

Air is air! (1) solution of oxygen gas dissolved in nitrogen gas. The oxygen in air is the (2) solute and nitrogen is the (3) solvent. Because oxygen gas dissolves in a solvent, oxygen gas is air's (4) soluble substance. A substance that does not dissolve is (5) insoluble. (6) Liquid solutions are the most common type of solutions. If one liquid is soluble in another liquid, such as acetic acid in water, the two liquids are (7) miscible. However, if one liquid is insoluble in another, the liquids are (8) immiscible.

Read about solvation in aqueous solutions in your textbook.

The diagram shows the hydration of solid sodium chloride to form an aqueous solution. Use the diagram to answer the following questions.



8. Hydration is solvation in which the solvent is water. What is solvation?

Solvation is the process of surrounding solute particles with solvent particles to form a solution.

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CHAPTER 15 STUDY GUIDE FOR CONTENT MASTERY

Section 15.1 continued

10. As sodium chloride dissolves in water, what happens to the sodium and chloride ions?

The sodium and chloride ions are separated and surrounded by the water molecules.

11. Explain the attraction of the water molecules around the sodium ions and chloride ions.

Because the sodium ion is positively charged, it attracts the negatively charged portion of the water molecule (the oxygen atom) and repels the positively charged portion of the water molecule (the hydrogen atoms). Because the chloride ion is negatively charged, it attracts the positively charged portion of the water molecule and repels the negatively charged portion.

12. How does the strength of the attraction between water molecules and sodium and chloride ions compare with the strength of the attraction between the sodium ions and chloride ions? How do you know?

The attraction between the water molecules and the sodium and chloride ions is greater than the attraction between the sodium and chloride ions. The greater strength of attraction between the water molecules and the ions is what causes the solvation process to occur.

13. List three ways that the rate of solvation may be increased.

stirring or shaking the solution, breaking the solute into smaller pieces, and heating the solvent

In your textbook, read about heat of solution, solubility, and factors that affect solubility.

For each statement below, write true or false.

True _____ 14. The overall energy change that occurs when a solution forms is called the heat of solution.

False _____ 15. Solubility is a measure of the maximum amount of solute that dissolves in a given amount of solvent at a specified temperature and pressure.

False _____ 16. Solvation continues as long as the solvation rate is less than the crystallization rate.

True _____ 17. In a saturated solution, solvation and crystallization are in equilibrium.

True _____ 18. Additional solute can be dissolved in an unsaturated solution.

True _____ 19. The solubility of a gas dissolved in a liquid decreases as the temperature of the solution increases.

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Gregory J. Privitera



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of the guide along with references to the appropriate textbook page and the textbook objective so that you can easily find where the material is presented Study tips in the Preface provide an overview of the most effective methods for learning and retaining information

Study Guide and Laboratory Manual for Physical Examination and Health Assessment, Canadian Edition- E-Book Carolyn Jarvis,Ann L. Eckhardt,2023-02-23 NEW Preface includes an overview of the NCSBN Clinical Judgement Measurement Model PN competencies and the REx PN NEW Clinical judgement exercises equip students for success on the Next Generation NCLEX NGN including questions with an increased focus on clinical judgement Updated content throughout corresponds to the 4th Canadian edition of the textbook and incorporates the latest research and evidence informed practice

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Designing Instruction for the Traditional, Adult, and Distance Learner: A New Engine for Technology-Based Teaching Tomei, Lawrence A.,2009-09-30 This book explores how technology impacts the process of devising instructional plans for adult students Provided by publisher

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How-to Guide for Active Learning Alice Fornari,Ann Poznanski,2021-05-21 This book focuses on large and small group educational settings and offers brief strategies to engage learners to assure active learning strategies are core to the learning environment The book opens with an introduction on active learning principles Each chapter follows with a specific description of a strategy written by authors who are experienced in using the strategy in a classroom environment with students The chapters are designed to be accessible and

practical for the reader to apply in their learning environments

Developing Learner-Centered Teaching Phyllis Blumberg, 2017-07-27 Developing Learner Centered Teaching offers a step by step plan for transforming any course from teacher centered to the more engaging learner centered model Filled with self assessments and worksheets that are based on each of the five practices identified in Maryellen Weimer's Learner Centered Teaching this groundbreaking book gives instructors faculty developers and instructional designers a practical and effective resource for putting the learner centered model into action

Exploring Management John R. Schermerhorn, Jr., Daniel G. Bachrach, 2017-10-16 Exploring Management supports teaching and learning of core management concepts by presenting material in a straightforward conversational style with a strong emphasis on application With a focus on currency high interest examples and pedagogy that encourages critical thinking and personal reflection Exploring Management is the perfect balance between what students need and what instructors want

The School Leader's Guide to Learner-Centered Education Barbara L. McCombs, Lynda Miller, 2009 The time is right and the need is profound for learner centered practices in schools McCombs and Miller describe the tools available and how a school can use them to make substantial changes that increase student performance Jean W Pierce Professor of Educational Psychology Northern Illinois University A new leadership paradigm for increasing collaboration and boosting achievement In today's educational environment school leaders must find new ways to create a positive school climate for staff and students while helping them meet rigorous testing and accountability requirements The School Leader's Guide to Learner Centered Education introduces a transformative leadership model for creating and sustaining a learner centered school environment Using core principles of educational psychology the authors examine how educators can support student growth development and achievement by focusing on key factors such as cognition metacognition motivation affect and individual differences Numerous reflective exercises assessment tools and real life success stories help readers Provide moral leadership for creating school change Incorporate learning standards without sacrificing student engagement Form learner centered support groups that utilize networking and technology Prepare students for the challenges of global citizenship Promote success for diverse populations and marginalized groups Designed to meet the needs of a rapidly evolving world this new leadership paradigm provides school principals superintendents and district leaders with the means to engage all members of the educational community in a lifelong process of growth improvement and empowerment

Environmentl Science Clark E. Adams, 2004-06 This study guide helps students identify the important concepts from the text and then provides them with review exercises study questions self check exercises and vocabulary review

Incorporating Applied Behavior Analysis into the General Education Classroom Jennifer Quigley, Mindy J. Cassano, Julie A. Ackerlund Brandt, 2023-08-28 This user friendly textbook addresses the relevance of Applied Behavior Analysis ABA within the general education setting Oftentimes ABA is thought of as only applicable to those with autism spectrum disorder and other developmental disorders In contrast the application of ABA principles is appropriate

for any socially significant problem such as those that arise daily in the general education setting This book will help educators through individual and group interventions decrease maladaptive behavior in their classrooms and to increase the focus on the curriculum This book not only summarizes the basics of ABA technologies and methodologies but provides readers with Materials tailored to be used with any student by general education practitioners teachers and clinicians Specific examples to ease the application of these technologies and curriculums Templates to be individualized to each reader s needs A focus on the functionality and applicability of these programs Written in layman s terms and broken down into categories of basic concepts and school wide implementation general educators will not only learn about ABA applications in education but will have clear examples of how to bridge the gap between understanding and application

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