

key

Molecular Geometry

Use the table to design for practice using the VSEPR model

Notes:

When you design your assignment, include an example, and including a new chemical compound, one of the groups. Include Lewis structures, molecular geometry, and bond angles. The table will be a reference to help you determine the molecular geometry, physical and chemical properties for the molecule. Be different for your assignment from the previous assignments.

Model 1 - Lewis Structures

Lewis Structures	Molecular Geometry	Bond Angle	Molecular Shape
	4 electron domains (4 bonding, 0 nonbonding)	109.5°	Tetrahedral
	4 electron domains (4 bonding, 0 nonbonding)	109.5°	Tetrahedral
	4 electron domains (4 bonding, 0 nonbonding)	109.5°	Tetrahedral
	4 electron domains (4 bonding, 0 nonbonding)	109.5°	Tetrahedral
	4 electron domains (4 bonding, 0 nonbonding)	109.5°	Tetrahedral
	4 electron domains (4 bonding, 0 nonbonding)	109.5°	Tetrahedral
	4 electron domains (4 bonding, 0 nonbonding)	109.5°	Tetrahedral
	4 electron domains (4 bonding, 0 nonbonding)	109.5°	Tetrahedral
	4 electron domains (4 bonding, 0 nonbonding)	109.5°	Tetrahedral

Molecular Geometry Answer Key Pogil

Robert R. Redfield



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