

1. What volume of hydrogen gas at STP is produced from the reaction of 50.0g of Mg and 75.0 grams of HCl? How much of the excess reagent is left over (in grams)?



Do a standard mass to mass problem starting with each reactant

50.0 g Mg	1 mol Mg 24.31 g Mg	1 mol H ₂ 1 mol Mg	22.4 L 1 mol H ₂	= 46.1 L H ₂
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75.0 g HCl	1 mol HCl 36.46 g HCl	1 mole H ₂ 2 mol HCl	22.4 L 1 mol H ₂	= 23.0 L H ₂
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HCl is the limiting reactant!!

Limiting Reactant Pogil Key

Raffaela Di Napoli

A red circular graphic with a gradient, appearing as a stylized arrow or a partial circle, located to the right of the author's name.

Limiting Reactant Pogil Key:

Adopting the Tune of Expression: An Emotional Symphony within **Limiting Reactant Pogil Key**

In some sort of taken by monitors and the ceaseless chatter of quick interaction, the melodic splendor and mental symphony developed by the published term usually diminish in to the back ground, eclipsed by the constant sound and disturbances that permeate our lives. However, nestled within the pages of **Limiting Reactant Pogil Key** a stunning fictional value overflowing with raw emotions, lies an immersive symphony waiting to be embraced. Constructed by a masterful musician of language, that charming masterpiece conducts readers on a mental trip, skillfully unraveling the hidden melodies and profound influence resonating within each cautiously crafted phrase. Within the depths with this poignant analysis, we shall explore the book is main harmonies, analyze its enthralling publishing design, and surrender ourselves to the profound resonance that echoes in the depths of readers souls.

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