

1. Fill in the chart below

Name of element	Atomic number	# of protons	# of electrons	Category of element (metal, nonmetal, metalloid)	Valence electrons (go to column #) (no. if larger than 8)	# of orbitals (period # = #)	Bohr model of the element Draw Beryllium & Neon in the space below
Beryllium	4	4	4	Metal	2	2	
Silicon	14	14	14	metalloid	4	3	
Neon	10	10	10	Non-metal	8	2	
Aluminum	13	13	13	metal	3	3	
Carbon	6	6	6	non-metal	4	2	

2. Orbitals are also called energy levels

3. Most of an atom's mass is found in the nucleus which is made up of protons & neutrons

4. ID the number of elements, molecules, & atoms in the compounds below

Compound	# of elements	Molecules	Element	atoms	Element	Atoms	Element	Atoms
$Fe(OH)_3$	3	7	Iron	7	Oxygen	14	Hydrogen	42
$Al_2(SO_4)_3$	3	4	Aluminum	12	Sulfur	12	Oxygen	48
Mg_3O_2	2	2	Magnesium	4	Chlorine	6		
$138 PO_4$	2	13	Phosphorus	39	Phosphorus	13	Oxygen	39
$MgSO_4$	3	1	Magnesium	1	Sulfur	1	Oxygen	4

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Table of Contents Houghton Unit 9 Benchmark Review Answers Science

1. Understanding the eBook Houghton Unit 9 Benchmark Review Answers Science
 - The Rise of Digital Reading Houghton Unit 9 Benchmark Review Answers Science
 - Advantages of eBooks Over Traditional Books
2. Identifying Houghton Unit 9 Benchmark Review Answers Science
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Houghton Unit 9 Benchmark Review Answers Science
 - User-Friendly Interface
4. Exploring eBook Recommendations from Houghton Unit 9 Benchmark Review Answers Science
 - Personalized Recommendations
 - Houghton Unit 9 Benchmark Review Answers Science User Reviews and Ratings
 - Houghton Unit 9 Benchmark Review Answers Science and Bestseller Lists
5. Accessing Houghton Unit 9 Benchmark Review Answers Science Free and Paid eBooks

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 - Houghton Unit 9 Benchmark Review Answers Science Compatibility with Devices
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- Adjustable Fonts and Text Sizes of Houghton Unit 9 Benchmark Review Answers Science
 - Highlighting and Note-Taking Houghton Unit 9 Benchmark Review Answers Science
 - Interactive Elements Houghton Unit 9 Benchmark Review Answers Science
8. Staying Engaged with Houghton Unit 9 Benchmark Review Answers Science
- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Houghton Unit 9 Benchmark Review Answers Science
9. Balancing eBooks and Physical Books Houghton Unit 9 Benchmark Review Answers Science
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Houghton Unit 9 Benchmark Review Answers Science
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Houghton Unit 9 Benchmark Review Answers Science
- Setting Reading Goals Houghton Unit 9 Benchmark Review Answers Science
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Houghton Unit 9 Benchmark Review Answers Science
- Fact-Checking eBook Content of Houghton Unit 9 Benchmark Review Answers Science
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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