# KEY!

# Chemistry 12 - Review of Significant Digits

TOTAL (28)

The rules for zeros in significant digits are as follows:

- a) All zeros between non-zero digits are significant.
- b) Zeros at the <u>beginning of a number</u> (eg. 0.0095) are NOT SIGNIFICANY! If the number 0.0095 was written in scientific notation, it would be: 9.5 x 10<sup>-3</sup>. The exponent is not counted as significant so this number has 2 significant digits.
- Zeros on the <u>right side</u> of a number (at the end) are significant if the DECIMAL POINT is shown.
  - eg) 50.00 has 4 significant digits 43.0 has 3 significant digits 20, has 2 significant digits 100, has 3 significant digits
- Zeros to the left of an UNDERSTOOD decimal point are NOT significant.
  - eg) 300 has I significant digit 10 000 has I significant digit 12 320 has 4 significant digits 420 has 2 significant digits
- Find the number of significant digits in each of the following measurements:

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# **Table of Contents Instructional Fair Worksheets Answer Key**

- 1. Understanding the eBook Instructional Fair Worksheets Answer Key
  - The Rise of Digital Reading Instructional Fair Worksheets Answer Key
  - Advantages of eBooks Over Traditional Books
- 2. Identifying Instructional Fair Worksheets Answer Key
  - Exploring Different Genres
  - o Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Instructional Fair Worksheets Answer Key
  - User-Friendly Interface
- 4. Exploring eBook Recommendations from Instructional Fair Worksheets Answer Key
  - Personalized Recommendations
  - Instructional Fair Worksheets Answer Key User Reviews and Ratings
  - Instructional Fair Worksheets Answer Key and Bestseller Lists

- 5. Accessing Instructional Fair Worksheets Answer Key Free and Paid eBooks
  - Instructional Fair Worksheets Answer Key Public Domain eBooks
  - Instructional Fair Worksheets Answer Key eBook Subscription Services
  - o Instructional Fair Worksheets Answer Key Budget-Friendly Options
- 6. Navigating Instructional Fair Worksheets Answer Key eBook Formats
  - ∘ ePub, PDF, MOBI, and More
  - Instructional Fair Worksheets Answer Key Compatibility with Devices
  - o Instructional Fair Worksheets Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Instructional Fair Worksheets Answer Key
  - Highlighting and Note-Taking Instructional Fair Worksheets Answer Key
  - Interactive Elements Instructional Fair Worksheets Answer Key
- 8. Staying Engaged with Instructional Fair Worksheets Answer Key
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Instructional Fair Worksheets Answer Key
- 9. Balancing eBooks and Physical Books Instructional Fair Worksheets Answer Key
  - $\circ$  Benefits of a Digital Library
  - Creating a Diverse Reading Collection Instructional Fair Worksheets Answer Key
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Instructional Fair Worksheets Answer Key
  - Setting Reading Goals Instructional Fair Worksheets Answer Key
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Instructional Fair Worksheets Answer Key
  - Fact-Checking eBook Content of Instructional Fair Worksheets Answer Key
  - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks
- 14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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