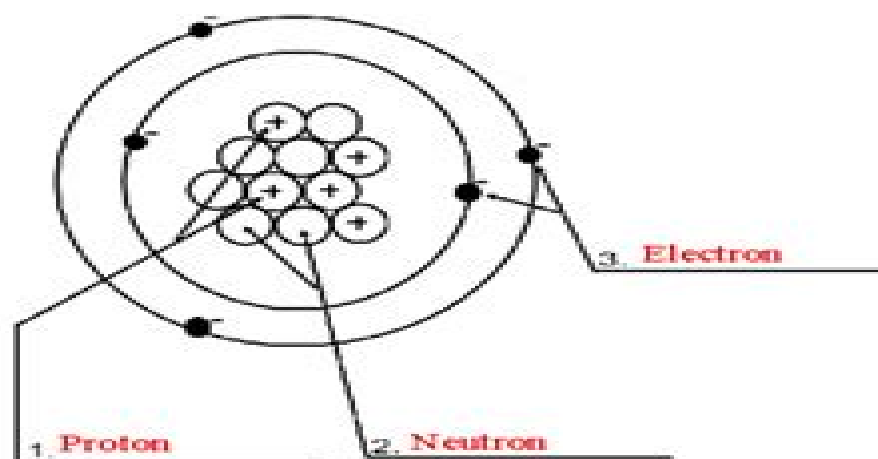


Atomic Structure Worksheet

Label the parts of an atom on the diagram below.



4. What type of charge does a proton have?

Positive (+1)

5. What type of charge does a neutron have?

Neutral (0)

6. What type of charge does an electron have?

Negative (-1)

7. Which two subatomic particles are located in the nucleus of an atom?

Protons and neutrons

8. If an atom has 35 protons in the nucleus, how many electrons will it have orbiting the nucleus?

35

9. What is the atomic number of the atom in the diagram above?

5

10. What is the atomic mass/mass number of the atom in the diagram above?

(11+11=22) *protons and neutrons added together) 22 is the atomic mass/mass number

11. How many protons are in the nucleus of an atom with an atomic number of 15?

15 *number of protons is the same as the atomic number

12. How many electrons are in the nucleus of an atom with an atomic number of 20?

20

13. How many neutrons are in the nucleus of an atom with an atomic number of 25?

(use Periodic Table for mass)

30 Neutrons. atomic mass: 54.938 ~ 55 atomic number: 25.....55-25=30

14. What is the mass number of an atom with 3 protons, 4 neutrons, and 3 electrons?

Mass number= 6.94. (Li-Lithium has an atomic number of 3 which means it has 3 protons.

15. How many neutrons are in the nucleus of an atom that has an atomic mass of 36 and an atomic number of 25?

Neutrons= 11 (atomic mass: 36 - 25 :atomic number)

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Camilla Rothe



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Table of Contents Instructional Fair Inc Atomic Structure Answer Key

1. Understanding the eBook Instructional Fair Inc Atomic Structure Answer Key
 - The Rise of Digital Reading Instructional Fair Inc Atomic Structure Answer Key
 - Advantages of eBooks Over Traditional Books
2. Identifying Instructional Fair Inc Atomic Structure Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Instructional Fair Inc Atomic Structure Answer Key
 - User-Friendly Interface
4. Exploring eBook Recommendations from Instructional Fair Inc Atomic Structure Answer Key
 - Personalized Recommendations
 - Instructional Fair Inc Atomic Structure Answer Key User Reviews and Ratings
 - Instructional Fair Inc Atomic Structure Answer Key and Bestseller Lists
5. Accessing Instructional Fair Inc Atomic Structure Answer Key Free and Paid eBooks
 - Instructional Fair Inc Atomic Structure Answer Key Public Domain eBooks
 - Instructional Fair Inc Atomic Structure Answer Key eBook Subscription Services
 - Instructional Fair Inc Atomic Structure Answer Key Budget-Friendly Options

6. Navigating Instructional Fair Inc Atomic Structure Answer Key eBook Formats
 - ePub, PDF, MOBI, and More
 - Instructional Fair Inc Atomic Structure Answer Key Compatibility with Devices
 - Instructional Fair Inc Atomic Structure Answer Key Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Instructional Fair Inc Atomic Structure Answer Key
 - Highlighting and Note-Taking Instructional Fair Inc Atomic Structure Answer Key
 - Interactive Elements Instructional Fair Inc Atomic Structure Answer Key
8. Staying Engaged with Instructional Fair Inc Atomic Structure Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Instructional Fair Inc Atomic Structure Answer Key
9. Balancing eBooks and Physical Books Instructional Fair Inc Atomic Structure Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Instructional Fair Inc Atomic Structure Answer Key
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Instructional Fair Inc Atomic Structure Answer Key
 - Setting Reading Goals Instructional Fair Inc Atomic Structure Answer Key
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Instructional Fair Inc Atomic Structure Answer Key
 - Fact-Checking eBook Content of Instructional Fair Inc Atomic Structure Answer Key
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

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