

Name _____

Molecular Geometries Lab Report Sheet

Formula	Lewis Structure	Bot electron domains/Wlp	Electron domain geometry	VSEPR Molecular geometry	Hybridization of central atom	Polar ?
CO_2	$\text{O}=\text{C}=\text{O}$ 24 valence 24 2 charge clouds	0	Linear 180°	Linear 	sp	NP
BF_3	$\text{F}-\text{B}-\text{F}$ 24 valence 32 3 charge clouds	0	Trigonal planar 120°		sp^2	NP
SnCl_2	$\text{Cl}-\text{Sn}-\text{Cl}$ 16 valence 24 2 charge clouds	3	Bent <120°		sp^2	P
BF_4^-	$\left[\text{F}-\text{B}-\text{F} \right]^-$ 24 valence 40 4 charge clouds	0	Tetrahedral 109.5°		sp^3	P
AsH_3	$\text{H}-\text{As}-\text{H}$ 8 valence 26 3 charge clouds	4	Trigonal pyramidal 109.5°		sp^3	P
H_2S	$\text{H}-\text{S}-\text{H}$ 8 valence 26 3 charge clouds	2	Bent 109.5°		sp^3	NP
NH_2^-	$\left[\text{H}-\text{N}-\text{H} \right]^-$ 8 valence 10 3 charge clouds	2	Bent 109.5°		sp^3	P
POF_3	$\text{F}-\text{P}-\text{Cl}$ 10 valence 32 4 charge clouds	0	Trigonal bipyramidal 120° + 90°		sp^3d	P

Lewis Structure And Molecular Models Lab Answers

Wolfgang Guggemos



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