

Question Number	Acceptable Answers	Reject	Mark
14(a)	<p>So that the phenol is used up / methyl orange is bleached before the rate changes (significantly)</p> <p>OR</p> <p>So that the phenol is used up / methyl orange is bleached during the initial rate period</p> <p>OR</p> <p>So that the concentration of bromide/bromate/reactants does not fall significantly before all the phenol is used up / the methyl orange is bleached</p> <p>OR</p> <p>Within this region/period/time the average rate of reaction approximates to the initial rate</p>	bromine	(1)

Question Number	Acceptable Answers	Reject	Mark
14(b)(i)	<p>So that only the <b>concentration</b> of bromide ions varies (significantly) during the course of the reaction / so that the <b>concentration</b> of the bromide ions is the limiting factor / so that the <b>concentration</b> of bromide ions is the only variable</p> <p>ALLOW</p> <p>So their <b>concentrations</b> / the <math>\text{BrO}_3^-</math> and <math>\text{H}^+</math> <b>concentrations</b> do not change</p> <p>OR</p> <p>So their <b>concentrations</b> / the <math>\text{BrO}_3^-</math> and <math>\text{H}^+</math> <b>concentrations</b> are not the limiting factor</p>		(1)

# English Past Paper Pixl

**RJ Shavelson**

A red circular graphic with a gradient, appearing as a stylized arrow or a partial circle, located to the right of the author's name.

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