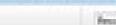


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Paul Laches Horseusin Chapter 14

Exercise 54 III) with feedback







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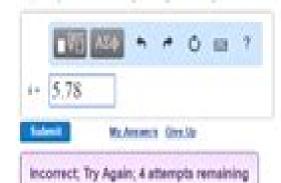
Exercise 14.90 with feedback

You may want to reference (<u>121, pages 525 - 542</u>) section 14.7 while completing this problem: A 9.16 m agreeus solution of an ionic compound with the formula MX has a freezing point of -2.5 °C.

Part A

Calculate the vanit Hoff factor (ii) for MX at this concentration.

Express your answer using two significant figures.





























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C Cleary

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Study Guide to Accompany Basics for Chemistry Martha Mackin, 2012-12-02 Study Guide to Accompany Basics for Chemistry is an 18 chapter text designed to be used with Basics for Chemistry textbook Each chapter contains Overview Topical Outline Skills and Common Mistakes which are all keyed to the textbook for easy cross reference The Overview section summarizes the content of the chapter and includes a comprehensive listing of terms a summary of general concepts and a list of numerical exercises while the Topical Outline provides the subtopic heads that carry the corresponding chapter and section numbers as they appear in the textbook The Fill in Multiple Choice are two sets of questions that include every concept and numerical exercise introduced in the chapter and the Skills section provides developed exercises to apply the new concepts in the chapter to particular examples The Common Mistakes section is designed to help avoid some of the errors that students make in their effort to learn chemistry while the Practical Test section includes matching and multiple choice questions that comprehensively cover almost every concept and numerical problem in the chapter After briefly dealing with an overview of chemistry this book goes on exploring the concept of matter energy measurement problem solving atom periodic table and chemical bonding These topics are followed by discussions on writing names and formulas of compounds chemical formulas and the mole chemical reactions calculations based on equations gases and the properties of a liquid The remaining chapters examine the solutions acids bases salts oxidation reduction reactions electrochemistry chemical kinetics and equilibrium and nuclear organic and biological chemistry This study guide will be of great value to chemistry teachers and students Organic Chemistry Marye Anne Fox, James K. Whitesell, 2004 Accompanying CD ROM has been enhanced with updated animated illustrations to accompany the presentations and Chem3D files for helpful structure visualization Page 4 of cover Introduction to Organic and Biological Chemistry Michael S. Matta, Antony C. Wilbraham, Dennis D. Stalev, 1996 Fundamentals of General, Organic, and Biological Chemistry John R. Holum, 1982 User friendly this comprehensive and up to date edition is organized according to the principle that life in both health and disease has a molecular basis Carefully worked example exercises enhanced by Analysis and Check sections enable students to examine and think about a problem not just answer it Includes numerous environmental applications *Introductory Chemistry* Charles H. Corwin, 2011 To help students learn chemical skills and concepts more effectively Introductory Chemistry Concepts and Critical Thinking Sixth Edition highlights the connection between key concepts and key problem solving skills through critical thinking Math and problem solving are covered early in the text Corwin builds your problem solving ability through innovative learning aids and technology formulated to meet your needs This revision retains all the strengths of the previous editions while adding emphasis on conceptual understanding and critical thinking **Introductory Chemistry for** Today Spencer L. Seager, Michael R. Slabaugh, 2004 Distinguished by its superior allied health focus and integration of technology Seager and Slabaugh's INTRODUCTORY CHEMISTRY FOR TODAY Fifth Edition continues to lead the market on

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