

## Gas Laws & SCUBA Diving

Read the accompanying article "Gas Laws and SCUBA Diving," ChemMatters, February 1993, pp.24.

Answer the following questions completely and upload in the assignment area.

1. What is gas? name and describe?
2. What does diving "30 meters below" and "not allow our bodies more than 100 lbs. of additional 30 meters below sea level?"
3. What parts of a diver's body are most affected by pressure changes?
4. What Boyle's Law?
5. What does a BCD (Buoyancy Compensator) do (why important)?
6. What would happen to a diver who ascends too rapidly without proper 30 meter stop? Explain in terms of Boyle's Law.
7. What Henry's Law?
8. What gas is associated with a warning bubble for this reason and why? Explain?
9. What is another reason for decompression sickness?
10. Explain how divers decompress by then a bubble follows equilibrium decompression tables.
11. What is nitrogen narcosis?
12. What is gas law equation and why is it important? (What does amount of gas and change in volume that total volume is higher? Explain).
13. What relationships between the temperature of gases and the solubility of a gas in a liquid or in a mixture of liquids? Explain.
14. What gas law does the BCD use to explain why it is important for a diver to have a full amount of air in the tank?

# Gas Laws And Scuba Diving Chemmatters Answers

**S Baum**



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