

Osmosis in Eggs **KEY**

Day I Questions:

1. Why are you placing the eggs in vinegar? What do you think will happen to the egg overnight?
Student answers will vary, but they may hypothesize that the vinegar will help remove the egg's shell based on seeing it bubble or based on previous experiences.

Day II Questions:

2. What happened to the mass of the eggs? Why do you think this happened?
The mass and circumference of the eggs increases. Student answers, as to why this happened, may vary but the correct answer is: because vinegar has more water than the inside of the egg, so water moves into the egg. Accept student answers that show thought and effort.
3. What do you think will happen to the egg in beaker #1?
Student answers will vary.
4. What do you think will happen to the egg in beaker #2?
Student answers will vary.

Day III Questions:

5. What happened to the egg in beaker #1? Why do you think this happened?
The egg put into water will continue to increase in mass and circumference. Student answers, as to why this happened, may vary but the correct answer is: there was more water outside the egg than inside the egg (and less solute outside than inside); therefore, water from the beaker moved into the egg through its membrane. Accept student answers that show thought and effort.
6. What happened to the egg in beaker #2? Why do you think this happened?
The egg in the corn syrup decreases in mass and circumference. It also appears shriveled. Student answers, as to why this happened, may vary but the correct answer is: there was more water inside the egg than in the corn syrup; therefore, water from the egg moved into the corn syrup.
7. The title of this lab is "Osmosis in Eggs". Based on that title and what you observed in the eggs, what do you think osmosis means?
Student answers will vary but should involve something to do with the movement of liquid or water across a membrane.

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