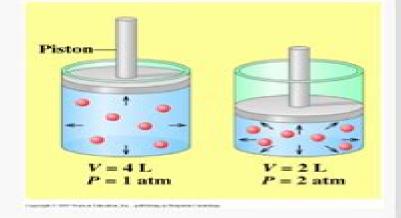
THE GAS LAWS

BOYLE'S LAW

- Robert Boyle
- at constant temperature, the volume of the gas increases as the pressure decreases. The volume of the gas decreases and the pressure increases.

$$P_1V_1 = P_2V_2$$



EXAMPLE:

A gas occupies 12.3 liters at a pressure of 40.0 mmHg. What
is the volume when the pressure is increased to 60.0 mmHg?

Gases Section 2 Review The Gas Laws

Bruce Fegley, Rose Osborne

Gases Section 2 Review The Gas Laws:

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with choices lettered A through E Material Tested The following chart summarizes the distribution of topics covered on the SAT II Chemistry Subject Test Topic Percentage Number of Questions Atomic however do not review too much at any one time Concentrate on one problem area at a time by reviewing the questions and explanations and by studying our review until you are confident you completely understand the material Keep track of your scores By doing so you will be able to gauge your progress and discover general weaknesses in particular sections You should carefully study the reviews that cover your areas of difficulty as this will build your skills in those areas TEST TAKING TIPS Although you may be unfamiliar with standardized tests such as the SAT II Chemistry Subject Test there are many ways to acquaint yourself with this type of examination and help alleviate your test taking anxieties Become comfortable with the format of the exam When you are practicing to take the SAT II Chemistry Subject Test simulate the conditions under which you will be taking the actual test Stay calm and pace yourself After simulating the test only a couple of times you will boost your chances of doing well and you will be able to sit down for the actual exam with much more confidence Know the directions and format for each section of the test Familiarizing yourself with the directions and format of the exam will not only save you time but will also ensure that you are familiar enough with the SAT II Chemistry Subject Test to avoid nervousness and the mistakes caused by being nervous Do your scratchwork in the margins of the test booklet You will not be given scrap paper during the exam and you may not perform scratchwork on your answer sheet Space is provided in your test booklet to do any necessary work or draw diagrams If you are unsure of an answer guess However if you do guess guess wisely Use the process of elimination by going through each answer to a question and ruling out as many of the answer choices as possible By eliminating three answer choices you give yourself a fifty fifty chance of answering correctly since there will only be two choices left from which to make your guess Mark your answers in the appropriate spaces on the answer sheet Fill in the oval that corresponds to your answer darkly completely and neatly You can change your answer but remember to completely erase your old answer Any stray lines or unnecessary marks may cause the machine to score your answer incorrectly When you have finished working on a section you may want to go back and check to make sure your answers correspond to the correct questions Marking one answer in the wrong space will throw off the rest of your test whether it is graded by machine or by hand You don't have to answer every question You are not penalized if you do not answer every question The only penalty results from answering a question incorrectly Try to use the guessing strategy but if you are truly stumped by a question remember that you do not have to answer it Work quickly and steadily You have a limited amount of time to work on each section so you need to work quickly and steadily Avoid focusing on one problem for too long Before the Test Make sure you know where your test center is well in advance of your test day so you do not get lost on the day of the test On the night before the test gather together the materials you will need the next day Your admission ticket Two forms of identification e g driver s license student identification card or current alien registration card Two No 2 pencils with erasers Directions to the test center A watch if

you wish but not one that makes noise as it may disturb other test takers On the day of the test you should wake up early after a good night's rest and have breakfast Dress comfortably so that you are not distracted by being too hot or too cold while taking the test Also plan to arrive at the test center early This will allow you to collect your thoughts and relax before the test and will also spare you the stress of being late If you arrive after the test begins you will not be admitted to the test center and you will not receive a refund During the Test When you arrive at the test center try to find a seat where you feel most comfortable Follow all the rules and instructions given by the test supervisor If you do not you risk being dismissed from the test and having your scores canceled Once all the test materials are passed out the test instructor will give you directions for filling out your answer sheet Fill this sheet out carefully since this information will appear on your score report After the Test When you have completed the SAT II Chemistry Subject Test you may hand in your test materials and leave Then go home and relax When Will I Receive My Score Report and What Will It Look Like You should receive your score report about five weeks after you take the test This report will include your scores percentile ranks and interpretive information Modern Chemistry Holt Rinehart & Winston, Holt, Rinehart and Winston Staff, 2001 Ralph Thomas Myers, 2004 Class 8-12 Chemistry Questions and Answers PDF Arshad Igbal, The Class 8 12 Chemistry Quiz Questions and Answers PDF Grade 8 12 Chemistry Competitive Exam Questions Chapter 1 15 Practice Tests Chemistry Textbook Questions for Beginners includes Questions to solve problems with hundreds of class guestions Class 8 12 Chemistry Questions and Answers PDF book covers basic concepts and analytical assessment tests Class 8 12 Chemistry Quiz PDF book helps to practice test questions from exam prep notes The Grade 8 12 Chemistry Quiz Questions and Answers PDF eBook includes Practice material with verbal quantitative and analytical past papers questions Class 8 12 Chemistry Questions and Answers PDF Free download chapter 1 a book to review textbook questions on chapters Molecular structure acids and bases atomic structure bonding chemical equations descriptive chemistry equilibrium systems gases laboratory liquids and solids mole concept oxidation reduction rates of reactions solutions thermochemistry Questions for high school and college revision questions Chemistry Interview Questions and Answers PDF Download free eBook s sample covers beginner's solved questions textbook's study notes to practice online tests The Grade 8 12 Chemistry Interview Questions Chapter 1 15 PDF book includes high school workbook questions to practice Questions for exam Chemistry Practice Tests a textbook s revision guide with chapters Questions for NEET MCAT GRE GMAT SAT ACT competitive exam Grade 8 12 Chemistry Questions Bank Chapter 1 15 PDF book covers problem solving exam tests from chemistry practical and textbook s chapters as Chapter 1 Molecular Structure Questions Chapter 2 Acids and Bases Questions Chapter 3 Atomic Structure Questions Chapter 4 Bonding Questions Chapter 5 Chemical Equations Questions Chapter 6 Descriptive Chemistry Questions Chapter 7 Equilibrium Systems Questions Chapter 8 Gases Questions Chapter 9 Laboratory Questions Chapter 10 Liquids and Solids Questions Chapter 11 Mole Concept Questions Chapter 12 Oxidation Reduction Questions Chapter 13 Rates of

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each Congressional session in the House of Representatives and the Senate Arranged in calendar order each volume includes the exact text of everything that was said and includes members remarks Code of Federal Regulations, Title 40, Protection of Environment, Pt. 81-84, Revised As of July 1 2012 .2012-10-26 **Code of Federal Regulations, Title** Thermodynamics Stephen R. 2009-10, **40, Protection of Environment, Pt. 81-84, Revised as of July 1, 2009** Turns, Laura L. Pauley, 2020-02-27 Presents an updated full color second edition on thermodynamics providing a structured approach to this subject and a wealth of new problems Code of Federal Regulations, Title 40, Protection of Environment, PT. 81-84, Revised as of July 1, 2010, 2010-10-08 Review of the Papers Presented in Section V, Railway & Steamship Refrigeration at the 2d International Congress of Refrigeration...,1911 **Nursing School Entrance Exams** Kaplan Nursing, 2016-08-02 Offers a comprehensive review of all tested material on major nursing school entrance assessments including the TEAS HESI PAX RN Kaplan and PSB RN exams Features 2 complete practice tests with detailed answer explanations Energy Abstracts for Policy Analysis ,1977-07 Physical Review ,1928 Vols for 1903 include Proceedings of the American Physical Society

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