

Topic 10: Polymer 10

**Question 1: Substitution and Addition Polymerization**

The reaction scheme below shows the mechanism of the reaction:

Identifying whether it is a free radical or ionic polymerization, explain the following components of the mechanism that occur before polymerization:

(a) Suggest the intermediate formed above the first substitution step using the appropriate structure. Write down:

(i) a type of free radical: alkyl radical. It has unpaired electrons with the "dot" next to the appropriate:

(ii) an:

(iii) an:

(iv) an:

(v) an:

(b) Explain the second phase of the polymerization mechanism:

(c) Draw and label the structure of the repeating unit formed by addition of monomers that polymerize via the mechanism below and suggest the mechanism:

(i) Electrophilic addition to form a carbocation:

(ii) An electrophilic addition to form a carbocation:

(iii) An addition to form a carbocation intermediate:

(iv) An addition to form a carbocation intermediate:

(v) An addition to form a carbocation:

(d) In the reaction scheme, the first step is the formation of a carbocation. How do you think this is achieved? Write down the following about the mechanism for the formation of the carbocation from the monomer:

(i) A free radical, possibly from the first step:

(ii) A free radical, possibly from the first step (from the first step):

(iii) A free radical, possibly from the first step (from the first step):

(iv) A free radical, possibly from the first step (from the first step):

(v) A free radical, possibly from the first step (from the first step):

**Question 2: The Free Radical Polymerization of Vinyl Monomers**

Write down:

(i) an addition to form a carbocation:

(ii) an addition to form a carbocation intermediate:

(iii) an addition to form a carbocation:

(iv) an addition to form a carbocation intermediate (from the first step):

(v) an addition to form a carbocation intermediate (from the first step):

# Microeconomics Multiple Choice 2013

**David Kirk**



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