

Empirical Formulas Practice Packet

Show all Calculations. Write final answers in the correct number of significant figures and with the correct units

1. A 7.546 g sample of an unknown compound was analyzed and found to contain only nitrogen and oxygen. If the mass of nitrogen present in the sample was 1.957 g, determine the empirical formula of the compound.
2. Serotonin is a compound that conducts nerve impulses in the brain. Serotonin has the following percent composition by mass: 68.2% carbon, 6.86% hydrogen, 15.9% nitrogen, and 9.08% oxygen.
 - a. What is its empirical formula?
 - b. What is its molecular formula if its molar mass is about 176 g/mol?
3. The koala dines exclusively on eucalyptus leaves. Its digestive system detoxifies the eucalyptus oil that can poison other animals. The chief constituent in eucalyptus oil is a substance called eucalyptol, which has the following mass percent composition: 77.87% carbon, 11.76% hydrogen, and the remainder is oxygen.
 - a. What is its empirical formula?
 - b. What is its molecular formula if its molar mass is about 154 g/mol?
4. Determine the molecular formulas for the following compounds given their empirical formulas and their molar masses: a) isoamyl formate: empirical formula= $\text{C}_3\text{H}_6\text{O}$, molar mass=116 g/mol; b) lactic acid: empirical formula= CH_2O , molar mass=90.1 g/mol; and c) opionic acid: empirical formula= $\text{C}_3\text{H}_2\text{O}$, molar mass=210.2 g/mol.

Empirical Formula Packet

**Barron's Educational Series, Gregory
Scott Hunter**

Empirical Formula Packet:

CliffsNotes Chemistry Practice Pack Charles Henrickson, 2010-02-08 About the Contents Pretest Helps you pinpoint where you need the most help Topic Area Reviews Measurement and Units of Measurement Matter Elements Compounds and Mixtures Atoms I The Basics Formulas and Names of Ionic Compounds Acids and Bases The Mole Elements and Compounds Percent Composition and Empirical and Molecular Formulas Chemical Reactions and Chemical Equations Calculations Using Balanced Equations Atoms II Atomic Structure and Periodic Properties Chemical Bonding The Formation of Compounds Gases and the Gas Laws The Forces between Molecules Solids and Liquids Solutions and Solution Composition Acids Bases and Neutralization Glossary Customized Full Length Exam Covers all subject areas Pretest that pinpoints what you need to study most Clear concise reviews of every topic Targeted example problems in every chapter with solutions and explanations Customized full length exam that adapts to your skill level **MCAT Study Review Notes -**

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Matsusaka,2019-10-16 The Fourth Edition of Powder Technology Handbook continues to serve as the comprehensive guide to powder technology and the fundamental engineering processes of particulate technology while incorporating significant advances in the field in the decade since publication of the previous edition The handbook offers a well rounded perspective on powder technologies in gas and liquid phases that extends from particles and powders to powder beds and from basic

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