

CHAPTER 5: ELECTRONS IN ATOMS

CHEMISTRY

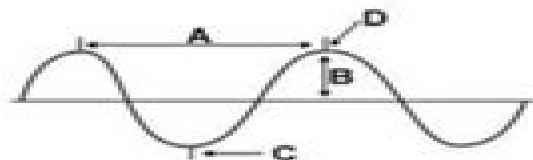
True/False

Indicate whether the statement is true or false.

- ____ 1. There are a total of 5 *d*-orbitals.
- ____ 2. The photoelectric effect describes the emission of electrons from a metal's surface when light at a specific frequency shines on it.
- ____ 3. Although the precise location of an electron cannot be determined, it is possible to determine the probability that an electron will occupy a certain region around the nucleus.
- ____ 4. In the visible light spectrum, red light has a greater frequency (energy) than violet light.
- ____ 5. $[\text{Kr}]5s^24d^{10}5p^1$ is the correct electron configuration for the element Bromine.
- ____ 6. The maximum number of allowed valence electrons in any atom is 8.
- ____ 7. Each element has its own unique emission spectrum.

Multiple Choice

Identify the choice that best completes the statement or answers the question.



- ____ 8. Which label identifies the amplitude of the wave shown?
 - a. A
 - b. B
 - c. C
 - d. D
- ____ 9. Which label identifies the wavelength of the wave shown?
 - a. A
 - b. B
 - c. C
 - d. D
- ____ 10. Which label identifies the trough of the wave shown?
 - a. A
 - b. B
 - c. C
 - d. D



- ____ 11. Which diagram shows a wave with the highest frequency?

Electrons In Atoms Chapter Answer Key

D Keegan



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