#### Kinetic Theory

- Gases are composed of \_\_\_\_\_\_

  barticles called \_\_\_\_\_\_.
- Gas molecules are in \_\_\_\_\_\_\_.
- All \_\_\_\_\_\_ between particles are \_\_\_\_\_\_
- The \_\_\_\_\_ of a gas display no \_\_\_\_\_ or \_\_\_\_ for one another.
- The \_\_\_\_\_\_ of the molecules is \_\_\_\_\_ to the \_\_\_\_\_ temperature of the gas.

#### Ideal Gas

Gas whose \_\_\_\_\_ conforms to the \_\_\_\_\_

—it is \_\_\_\_\_.

#### Gas Pressure:

Pressure = ----

Atmospheric Pressure - the \_\_\_\_\_ the earth's \_\_\_\_\_ exerts due to its

Barometer:

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Robert R. Redfield

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