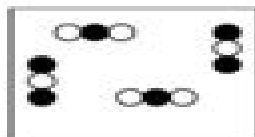


Elements, Compounds and Mixtures

1. Classify each of the following as elements (E), compounds (C) or Mixtures (M). Write the letter X if it is none of these.

<u>E</u> Diamond (C)	<u>C</u> Water (H_2O)	<u>C</u> Dry Ice (CO_2)
<u>C</u> Sugar ($C_6H_{12}O_6$)	<u>C</u> Alcohol (CH_3OH)	<u>C</u> Baking Soda ($NaHCO_3$)
<u>M</u> Milk	<u>M</u> Pail of Garbage	<u>E</u> Titanium (Ti)
<u>M</u> Air	<u>C</u> Ammonia (NH_3)	<u>E</u> Iron (Fe)
<u>C</u> Sulfuric Acid (H_2SO_4)	<u>C</u> Salt ($NaCl$)	<u>M</u> Popcorn
<u>M</u> Gasoline	<u>X</u> Energy	<u>E</u> Gold (Au)
<u>E</u> Krypton (Kr)	<u>M</u> Wood	<u>X</u> Electricity
<u>E</u> Bismuth (Bi)	<u>M</u> Bronze	<u>M</u> A dog
<u>E</u> Uranium (U)	<u>M</u> Ink	<u>M</u> Concrete

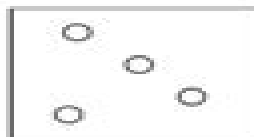
2. Match each diagram with its correct description. Diagrams will be used once.



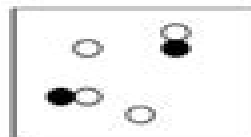
A



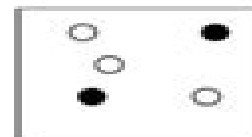
B



C



D



E

- C Pure Element – only one type of atom present.
- E Mixture of two elements – two types of uncombined atoms present.
- B Pure compound – only one type of compound present.
- A Mixture of two compounds – two types of compounds present.
- D Mixture of a compound and an element.

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Raffaella Di Napoli



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