

Molar Concentration Worksheet

Name: _____

Complete all questions including units in your calculations. Each question is worth 4 marks.

You may use Google to find the molar masses (ex. type 'molar mass HCl = ' and it will give you the answer).

1. Every 5.5 L of battery acid contains 490.0 g of sulfuric acid, H_2SO_4 . Calculate the concentration of sulfuric acid in the battery acid in mol/L.
2. Lily added 35 grams of sodium carbonate to a volume of 3,400 mL. What is the concentration of her solution in mol/L? (answer 0.097 mol/L)
3. If Jaclyn adds water to 100 mL of a 0.15 M NaOH solution until the final volume is 150 mL, what will the molarity (concentration in mol/L) of the diluted solution be?
4. Explain how Savannah could make 450 mL of a 0.25 M calcium chloride solution in the lab from solid calcium chloride if the only volumetric flask she has is a 500 mL volumetric flask. Show all calculations.

General Chemistry Unit 11 Worksheet 4 Molar Concentration

Judd E. Hollander



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Chemistry Carson-Dellosa Publishing, 2015-03-16 Chemistry for grades 9 to 12 is designed to aid in the review and practice of chemistry topics Chemistry covers topics such as metrics and measurements matter atomic structure bonds compounds chemical equations molarity and acids and bases The book includes realistic diagrams and engaging activities to support practice in all areas of chemistry The 100 Series science books span grades 5 to 12 The activities in each book reinforce essential science skill practice in the areas of life science physical science and earth science The books include engaging grade appropriate activities and clear thumbnail answer keys Each book has 128 pages and 100 pages or more of reproducible content to help students review and reinforce essential skills in individual science topics The series will be aligned to current science standards **Basics of Analytical Chemistry and Chemical Equilibria** Brian M.

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