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Ions and Isotopes Worksheet

Symbol	Name of the element	Atomic number	Protons	Neutrons	Electrons	Atomic Mass	Charge
O -17	Oxygen	8	8	.9	8	17	0
Xe - 131	Xenon	54	54	77	54	131	0
²⁰⁷ Pb	Lead	82	82	125	82	207	0
Nata	Sodium ion	11	11	12	10	2.3	+1
Sr	Strontium	38	38	50	.38	88	0
Al ³ *	Aluminum	13	13	14	10	27	+3
239U	Uranium	92	92	146	92	2.38	0
²⁵ As -3	Arsenic	33	33	39	36	75	-3
32S -2	Sulfide ion	16	16	16	18	32	-2
Na	Sodium	11	11	12	11	2.3	0
N3	Nitride ion	7	7	7	10	14	-3
53/Cu	Copper	29	29	36	29	65	0
¹²⁰ Hg	Mercury isotope	80	80	40	80	120	0
Ag	silver	47	47	61	47	108	0
Zr	zirconium	40	40	51	91	40	0
Cr	chromium	24	24	28	24	56	0
26.0 104 Rdf	rutherfordiu	104	104	158	104	261	0

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K Morrison

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Isotope User's Guide F. E. McKinney, P. S. Baker, S. A. Reynolds, 1969

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