

Second Semester Final Exam

Multiple Choice: Please mark the answer to the multiple choice questions on your bubble sheet.

- 1) The molar mass of magnesium cyanide, $\text{Mg}(\text{CN})_2$, is approximately
 - A) 50.3 g
 - B) 64.3 g
 - C) 76.3 g
 - D) 100.6 g
- 2) The total number of atoms in $(\text{NH}_4)_3\text{BO}_3$ is
 - A) 12
 - B) 17
 - C) 19
- 3) What is the mass in grams of 5.90 mol C_8H_{18} ?
 - A) 0.0512 g
 - B) 389 g
 - C) 673 g
 - D) 19.4 g
- 4) How many atoms are there in 1 mole of helium?
 - A) 6.02×10^{23} atoms
 - B) 22.4 atoms
 - C) 4.0 atoms
 - D) 0.0821 atoms
- 5) The percent composition of silver (Ag) in AgNO_3 is
 - A) 0.625 %
 - B) 1.57 %
 - C) 62.5 %
 - D) 157 %
 - E) 169.9 %
- 6) Which of the following is an empirical formula?
 - A) C_2H_4
 - B) C_2H_6
 - C) CH_4
 - D) C_3H_6
- 7) Which of the following could be a molecular formula for the empirical formula CH_2 ?
 - A) C_3H_6
 - B) C_6H_3
 - C) CH
 - D) CH_4
- 8) What is the coefficient in front of the O_2 when the following reaction is correctly balanced?
$$\text{C}_3\text{H}_8 + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$$
 - A) 1
 - B) 2
 - C) 3
 - D) 4
 - E) 5

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PT Brinkman

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