

1997

22. Avogadro's Hypothesis:

EQUAL VOLUMES at the same Temp & Pressure
CONTAIN EQUAL #'s of molecules.

B

#3 = Volumes, = Temp = moles $\therefore$ = pressure

#1 NOT TRUE

DIFF MASS = DIFF VELOCITIES

#2 NOT TRUESAME MOLES BUT DIFF. MASS = DIFF DENSITIES

1996

20.

ΔT	$\frac{288 K}{15^\circ C}$	$\rightarrow$	$\frac{298 K}{25^\circ C}$	$T \uparrow$	$V \uparrow$
ΔP	1.5	$\rightarrow$	1.0 atm	$P \downarrow$	$V \uparrow$

$$V \propto \frac{298}{288} \times \frac{1.5}{1.0} = V = 1.6$$

B

1995 16.

$$d = \frac{P_{MM}}{RT}$$

density of mm

mm less than Ar (39.95)

I. $CO_2 = 44$ II. $Cl_2 = 71$ (diatomic)!III. $O_2 = 32$ only

B

$$PV = nRT$$

$$P \cancel{V} \frac{g}{mm} = RT$$

$$d = \frac{g}{V} = \frac{P_{MM}}{RT}$$

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SA Dillow



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