# Polyatomic Ions Chart

Formula	Name	Formula	Name
NH4.	Ammonium	CrO <sub>4</sub> -2	Chromate
NH <sub>3</sub>	Ammonia	Cr2O7-2	Dichromate
C2H3O2"	Acetate	MnO <sub>4</sub> "	Permanganate
CH₃COO	Acetate	MnO <sub>4</sub> -≥	Manganate
CN	Cyanide	NO <sub>2</sub>	Nitrite
CO3-2	Carbonate	NO <sub>3</sub>	Nitrate
HCO3	Bicarbonate	OH.	Hydroxide
C2O4 2	Oxalate	PO4-3	Phosphate
CIO.	Hypochlorite	SCN"	Thiocyanate
CIO₂	Chlorite	Fe(CN)6-3	Ferricyanide
CIO3	Chlorate	5O₃ <sup>-2</sup>	Sulfite
CIO <sub>4</sub>	Perchlorate	50 <sub>4</sub> -2	Sulfate
5₂O₃-²	Thiosulfate	HSO₄	Hydrogen sulfate
BrO	Hypobromite	IO <sub>3</sub>	Iodate
AsO2-3	Arsenite	SeO <sub>4</sub> -2	Selenate
BrO <sub>3</sub>	Bromate	H5O₃ <sup>-</sup>	Hydrogen sulfite

## Commonly Used Multivalent Metals

Symbol	Name	-ous	-ic
Fe (Ferrum)	Iron	+2	+3
Pb (Plumbum)	Lead	+2	+4
Sn (Stannum)	Tin	+2	+4
Hg (Hydrargyum)	Mercury	+1	+2
Cu (Cuprum)	Copper	+1	+2

### Prefixes

Mono=one	Hexa = six	
Di = two	Hepta = seven	
Tri = three	Octa = eight	
Tetra = four	Nona = nine	
Penta = five	Deca = ten	

# **Flinn Scientific Polyatomic Ions**

**Rachel S Tattersall** 

### Flinn Scientific Polyatomic Ions:

Physics Briefs ,1981 Applied Science & Technology Index ,1979 Scientific and Technical Aerospace Reports ,1973 Coatings on Glass H. Pulker,H.K. Pulker,1999-03-29 This is the second revised edition of a book that has already proved invaluable to a wide range of readers Written by a scientist for scientists and technical people it goes beyond the subject matter indicated by the title filling the gap which previously existed in the available technical literature It includes a wealth of information for physicists chemists and engineers who need to know more about thin films for research purposes or who want to use this special form of solid material to achieve a variety of application oriented goals Insdoc List ,1962

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important information about a large number of semiconductors **Investigations Into the Origins of Polyatomic Ions in Inductively Coupled Plasma-mass Spectrometry**, 2010 An inductively coupled plasma mass spectrometer ICP MS is an elemental analytical instrument capable of determining nearly all elements in the periodic table at limits of detection in the parts per quadrillion and with a linear analytical range over 8 10 orders of magnitude Three concentric quartz tubes make up the plasma torch Argon gas is spiraled through the outer tube and generates the plasma powered by a looped load coil operating at 27 1 or 40 6 MHz The argon flow of the middle channel is used to keep the plasma above the innermost tube through which solid or aqueous sample is carried in a third argon stream A sample is progressively desolvated atomized and ionized The torch is operated at atmospheric pressure To reach the reduced pressures of mass spectrometers ions are extracted through a series of two approximately one millimeter wide circular apertures set in water cooled metal cones The space between the cones is evacuated to approximately one torr. The space behind the second cone is pumped down to or near to the pressure needed for the mass spectrometer MS The first cone called the sampler is placed directly in the plasma plume and its position is adjusted to the point where atomic ions are most abundant The hot plasma gas expands through the sampler orifice and in this expansion is placed the second cone called the skimmer After the skimmer traditional MS designs are employed i e quadrupoles magnetic sectors time of flight ICP MS is the leading trace element analysis technique One of its weaknesses are polyatomic ions. This dissertation has added to the fundamental understanding of some of these polyatomic ions their origins and behavior Although mainly continuing the work of others certain novel approaches have been introduced here Chapter 2 includes the first reported efforts to include high temperature corrections to the partition functions of the polyatomic ions in ICP MS This and other objections to preceeding papers in this area were addressed Errors in the measured T sub gas values were found for given errors in the experimental and spectroscopic values The ionization energy of the neutral polyatomic ion was included in calculations to prove the validity of ignoring more complicated equilibria Work was begun on the question of agreement between kinetics of the plasma and interface and the increase and depletion seen in certain polyatomic ions This dissertation was also the first to report day to day ranges for T sub gas values and to use a statistical test to compare different operating conditions. This will help guide comparisons of previous and future work Chapter 4 was the first attempt to include the excited electronic state 2 in the partition function of ArO as well as the first to address the different dissociation products of the ground and first electronic levels of ArO Chapter 5 reports an interesting source of memory in ICP MS that could affect mathematical corrections for polyatomic ions For future work on these topics I suggest the following experiments and investigations Clearly not an extensive list they are instead the first topics curiosity brings to mind 1 Measurement of T sub gas values when using the flow injection technique of Appendix B It was believed that there was a fundamental difference in the plasma when the auto sampler was used versus a continuous injection Is this reflected in T sub gas values 2 The work of Chapter 3 can be expanded and supplemented with more trials

new cone materials i e copper stainless steel and more cone geometries Some of this equipment is already present in the laboratory others could be purchased or made 3 T sub gas values from Chapter 3 could be correlated with instrument pressures during the experiment Pressures after the skimmer cone were recorded for many days but have yet to be collated with the measured T sub gas values 4 The work in Chapter 5 could be expanded to include more metals Does the curious correlation between measured T sub gas and element boiling point persist 5 Investigate non linear correlations to T sub gas values of the MO memory in Chapter 5 Temperatures along the skimmer walls are not a linear gradient Ring deposits have been observed on the cone and photographs of the interface show light intensities shaping a sort of tailing peak along the outside skimmer wall Is there a physical property of the metals or metal oxides that would give this peak with the T sub gas values 6 Chemical state speciation of the metal deposits on the skimmers of Chapter 5 There may be a more logical correlation between Tgas and a physical property of the deposit ing chemical if all the metals do not deposit in the same form 7 A collaboration with our computational collegues would be most welcome Newer calculations for ArO and RuO would be very helpful Investigations Into the Origins of Polyatomic Ions in Inductively Coupled Plasma-mass Spectrometry Sally M. McIntyre, 2010 Fragmentation of Diatomic and Polyatomic Ions in the Gas Phase Christopher John Proctor, 1981

Unveiling the Magic of Words: A Overview of "Flinn Scientific Polyatomic Ions"

In a world defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their capability to kindle emotions, provoke contemplation, and ignite transformative change is truly aweinspiring. Enter the realm of "Flinn Scientific Polyatomic Ions," a mesmerizing literary masterpiece penned by way of a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve into the book is central themes, examine its distinctive writing style, and assess its profound effect on the souls of its readers.

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