

GUIDED READING WITH CHUNKED QUESTIONS THROUGHOUT & POST-READING PRACTICE

ATOMS

1 PARTS OF AN ATOM


1.1. Atoms are made up of three types of particles: protons, neutrons and electrons. Protons and neutrons are found in the nucleus, which is at the center of the atom. Electrons are found outside the nucleus, in regions called electron shells.

1.2. The nucleus is made up of protons and neutrons. Protons have a positive charge, and neutrons have no charge. Electrons have a negative charge.

1.3. The electron shells are arranged in layers around the nucleus. The innermost shell is the closest to the nucleus, and the outermost shell is the furthest from the nucleus.

1.4. The number of protons in an atom is called the atomic number. The number of neutrons in an atom is called the mass number. The number of electrons in an atom is called the electron number.

1.5. The atomic number, mass number, and electron number of an atom can be used to identify the element.



Particle	Charge	Location
Proton	Positive	Nucleus
Neutron	Neutral	Nucleus
Electron	Negative	Electron shells

- ✓ SUBATOMIC PARTICLES
- ✓ ATOMIC MASS
- ✓ ATOMIC NUMBER
- ✓ ELECTRON SHELLS
- ✓ DRAWING DIAGRAMS

ATOMS

2 ELECTRON SHELLS

2.1. Electrons are arranged in shells around the nucleus. The innermost shell is the closest to the nucleus, and the outermost shell is the furthest from the nucleus.

2.2. The number of electrons in each shell is limited. The first shell can hold up to 2 electrons, the second shell can hold up to 8 electrons, and the third shell can hold up to 18 electrons.

2.3. The outermost shell of an atom is called the valence shell. The number of electrons in the valence shell determines the chemical properties of the element.

2.4. The number of protons in an atom is equal to the number of electrons in an atom. This is because the atom is electrically neutral.

2.5. The number of protons in an atom is equal to the atomic number.

3 ATOMS PRACTICE





3.1. Draw the structure of an atom with 11 protons, 11 electrons, and 12 neutrons.

3.2. Draw the structure of an atom with 17 protons, 17 electrons, and 18 neutrons.

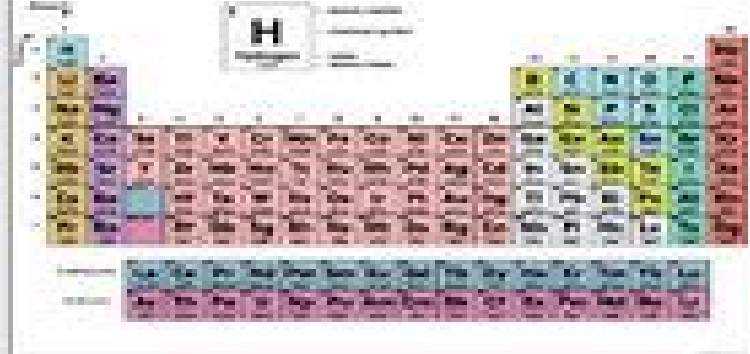
3.3. Draw the structure of an atom with 12 protons, 12 electrons, and 12 neutrons.

3.4. Draw the structure of an atom with 15 protons, 15 electrons, and 15 neutrons.

3.5. Draw the structure of an atom with 18 protons, 18 electrons, and 18 neutrons.

PERIODIC TABLE OF ELEMENTS



Periodic Table &
Answer Key Included!



Electrons In Atoms Guided Reading 5

Albert A Gayle



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