Learning Target: Use isotope notation to determine: element name/symbol, atomic number, number of electrons, number of neutrons, number of protons, mass number, atomic number, atomic mass. Isotope Notation:

- 1. Here are three isotopes of an element: 6^{12} C 6^{13} C 6^{14} C
 - a. The element is: ____earbon_____
 - b. The number 6 refers to the ____atomic number____

 - e. How many protons and neutrons are in the second isotope? P=6 N=7
 - f. How many protons and neutrons are in the third isotope? __P=6 N=7____
- 2. Complete the following chart:

Isotope name	atomic#	mass #	# of protons	# of neutrons	# of electrons	Isotope Notation
uranium-235	92	235	92	143	92	235 92-U
uranium-238	92	238	92	146	92	238 92- U
boron-10	5	10	5	5	5	*** **B
boron-11	5	11	5	6	5	sB

3.

Element	Symbol	Atomic Number	Number of electrons	Number of Neutrons	Mass number	Isotope Name	Isotope Notation
Helium	He	2	2	2	4	Helium-4	₄ ₂He
Titanium	Ti	22	22	28	50	Titanium-50	50 22 T/
Tantalum	Ta	73	73	108	181	Tantalum- 181	23 Ta
Gallium	Ga	31	31	39	70	Gallium-70	³⁰ 31 Ga
Carbon	C	6	6	7	13	Carbon-13	13C
Radium	Ra	88	88	218	226	Radium-226	seRa
Bismuth	Bi	83	83	127	210	Bismuth-210	216 83 B i

Isotopes And Ions Worksheet Answers

SA Adler

Isotopes And Ions Worksheet Answers:

Isotopes And Ions Worksheet Answers Book Review: Unveiling the Magic of Language

In a digital era where connections and knowledge reign supreme, the enchanting power of language has be apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "Isotopes And Ions Worksheet Answers," compiled by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we will delve into the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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