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Models by the same editors who are two of the top scientists working on this topic each with extensive experience and important connections within the community *How Chemical Bonds Form and Chemical Reactions Proceed* V. I. Gantmakher, Yuri V. Gankin, 1998 **Chemical Bonding at Surfaces and Interfaces** Anders Nilsson, Lars G.M. Pettersson, Jens Norskov, 2011-08-11 Molecular surface science has made enormous progress in the past 30 years The development can be characterized by a revolution in fundamental knowledge obtained from simple model systems and by an explosion in the number of experimental techniques The last 10 years has seen an equally rapid development of quantum mechanical modeling of surface processes using Density Functional Theory DFT Chemical Bonding at Surfaces and Interfaces focuses on phenomena and concepts rather than on experimental or theoretical techniques The aim is to provide the common basis for describing the interaction of atoms and molecules with surfaces and this to be used very broadly in science and technology The book begins with an overview of structural information on surface adsorbates and discusses the structure of a number of important chemisorption systems Chapter 2 describes in detail the chemical bond between atoms or molecules and a metal surface in the observed surface structures A detailed description of experimental information on the dynamics of bond formation and bond breaking at surfaces make up Chapter 3 Followed by an in depth analysis of aspects of heterogeneous catalysis based on the d band model In Chapter 5 adsorption and chemistry on the enormously important Si and Ge semiconductor surfaces are covered In the remaining two Chapters the book moves on from solid gas interfaces and looks at solid liquid interface processes In the final chapter an overview is given of the environmentally important chemical processes occurring on mineral and oxide surfaces in contact with water and electrolytes Gives examples of how modern theoretical DFT techniques can be used to design heterogeneous catalysts This book suits the rapid introduction of methods and concepts from surface science into a broad range of scientific disciplines where the interaction between a solid and the surrounding gas or liquid phase is an essential component Shows how insight into chemical bonding at surfaces can be applied to a range of scientific problems in heterogeneous catalysis electrochemistry environmental science and semiconductor processing Provides both the fundamental perspective and an overview of chemical bonding in terms of structure electronic structure and dynamics of bond rearrangements at surfaces *Lectures On Chemical Bonding And Quantum Chemistry* Sambhu N Datta, 2019-02-25 The concept of a chemical bond evolved from a variety of experimental observations It became useful to understand at times even predict the molecular structure reactivity and mechanism of chemical reactions Every aspect of the concept of bonding received a quantitative interpretation from the advent of quantum mechanics and its application to chemistry In Lectures on Chemical Bonding and Quantum Chemistry the reader will find a comprehensive discourse on the basic interpretation of the chemical bond as well as current understanding in terms of a dancing molecule that not only travels rotates and pulsates around an equilibrium molecular structure but also interacts and collides with other molecules thereby transferring linear and angular momentum characteristics and adjusting total energies

One will also find a thorough survey of quantum mechanical methodologies for calculation of molecular characteristics in specific states and their changes under spectroscopic transitions tunneling electron and proton transfer phenomena and so on. Guides to more advanced levels of theory are also provided.

Chemical Bonds Harry B. Gray, 1994-12-05 This profusely illustrated book by a world renowned chemist and award winning chemistry teacher provides science students with an introduction to atomic and molecular structure and bonding. This is a reprint of a book first published by Benjamin Cummings 1973.

Chemical Bonding and Molecular Geometry Ronald James Gillespie, Ronald J. Gillespie, Paul L. A. Popelier, 2001 Ideal for undergraduate and first year graduate courses in chemical bonding. Chemical Bonding and Molecular Geometry From Lewis to Electron Densities can also be used in inorganic chemistry courses. Authored by Ronald Gillespie a world class chemist and expert on chemical bonding and Paul Popelier of the University of Manchester Institute of Science and Technology this text provides students with a comprehensive and detailed introduction to the principal models and theories of chemical bonding and geometry. It also serves as a useful resource and an up to date introduction to modern developments in the field for instructors teaching chemical bonding at any level. Features Shows students how the concept of the chemical bond has developed from its earliest days through Lewis's brilliant concept of the electron pair bond and up to the present day. Presents a novel non traditional approach that emphasizes the importance of the Pauli principle as a basis for understanding bonding. Begins with the fundamental classical concepts and proceeds through orbital models to recent ideas based on the analysis of electron densities which help to clarify and emphasize many of the limitations of earlier models. Provides a thorough and up to date treatment of the well known valence shell electron pair VSEPR model which was first formulated and developed by author Ronald Gillespie and the more recent ligand close packing LCP model. Presents a unique pictorial and nonmathematical discussion of the analysis of electron density distributions using the atoms in molecules AIM theory. Emphasizes the relationships between these various models giving examples of their uses limitations and comparative advantages and disadvantages.

Chemical Bonding Kok Koon Leong, 2021-05-03 This book introduces the principles behind chemical bonding to teenagers between the ages of fifteen to seventeen. Topics covered include ionic bonding covalent bonding and metallic bonding.

Chemical Bonding Mark J. Winter, 2023 Chemical bonding which underlies the structure and reactivity of all molecules is explained beautifully clearly and succinctly in this primer making it the perfect text to introduce students to this fundamental aspect of chemistry.

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