

### Example 1

Calculate the  $K_a$  value of a 0.03 M aqueous solution of hydrofluoric acid (HF) with a pH of 2.75.



	HF	H <sub>2</sub> O	→	F <sup>-</sup>	H <sub>3</sub> O <sup>+</sup>
I	0.03	~		0	0
C	-x			+x	+x
E	0.03-x			+x	+x

pH = 2.75

$$[\text{H}_3\text{O}^{+}] = 10^{-2.75} = 1.8 \times 10^{-3} \text{ M}$$

$$[\text{F}^{-}] = 1.8 \times 10^{-3} \text{ M}$$

$$K_a = \frac{[\text{products}]}{[\text{reactants}]} = \frac{[\text{F}^{-}][\text{H}_3\text{O}^{+}]}{[\text{HF}]}$$

$$K_a = \frac{(x)(x)}{(0.03-x)} = \frac{x^2}{(0.03-x)}$$

$$K_a = \frac{(1.8 \times 10^{-3})^2}{(0.03 - (1.8 \times 10^{-3}))} = 1 \times 10^{-4}$$

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Multiple Representations in Chemical Education John K. Gilbert, David Treagust, 2009-02-28 Chemistry seeks to provide qualitative and quantitative explanations for the observed behaviour of elements and their compounds Doing so

involves making use of three types of representation the macro the empirical properties of substances the sub micro the natures of the entities giving rise to those properties and the symbolic the number of entities involved in any changes that take place Although understanding this triplet relationship is a key aspect of chemical education there is considerable evidence that students find great difficulty in achieving mastery of the ideas involved In bringing together the work of leading chemistry educators who are researching the triplet relationship at the secondary and university levels the book discusses the learning involved the problems that students encounter and successful approaches to teaching Based on the reported research the editors argue for a coherent model for understanding the triplet relationship in chemical education

*Powerful Ideas of Science and How to Teach Them* Jasper Green,2020-07-19 A bullet dropped and a bullet fired from a gun will reach the ground at the same time Plants get the majority of their mass from the air around them not the soil beneath them A smartphone is made from more elements than you Every day science teachers get the opportunity to blow students minds with counter intuitive crazy ideas like these But getting students to understand and remember the science that explains these observations is complex To help this book explores how to plan and teach science lessons so that students and teachers are thinking about the right things that is the scientific ideas themselves It introduces you to 13 powerful ideas of science that have the ability to transform how young people see themselves and the world around them Each chapter tells the story of one powerful idea and how to teach it alongside examples and non examples from biology chemistry and physics to show what great science teaching might look like and why Drawing on evidence about how students learn from cognitive science and research from science education the book takes you on a journey of how to plan and teach science lessons so students acquire scientific ideas in meaningful ways Emphasising the important relationship between curriculum pedagogy and the subject itself this exciting book will help you teach in a way that captivates and motivates students allowing them to share in the delight and wonder of the explanatory power of science

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theory to examples from the chemistry classroom Key sentences tasks for self assessment and suggestions for further reading are also included The book is focused on many different issues a teacher of chemistry is concerned with The chapters provide contemporary discussions of the chemistry curriculum objectives and assessment motivation learning difficulties linguistic issues practical work student active pedagogies ICT informal learning continuous professional development and teaching chemistry in developing environments This book with contributions from many of the world s top experts in chemistry education is a major publication offering something that has not previously been available Within this single volume chemistry teachers teacher educators and prospective teachers will find information and advice relating to key issues in teaching such as the curriculum assessment and so forth but contextualised in terms of the specifics of teaching and learning of chemistry and drawing upon the extensive research in the field Moreover the book is written in a scholarly style with extensive citations to the literature thus providing an excellent starting point for teachers and research students undertaking scholarly studies in chemistry education whilst at the same time offering insight and practical advice to support the planning of effective chemistry teaching This book should be considered essential reading for those preparing for chemistry teaching and will be an important addition to the libraries of all concerned with chemical education Dr Keith S Taber University of Cambridge Editor Chemistry Education Research and Practice The highly regarded collection of authors in this book fills a critical void by providing an essential resource for teachers of chemistry to enhance pedagogical content knowledge for teaching modern chemistry Through clever orchestration of examples and theory and with carefully framed guiding questions the book equips teachers to act on the relevance of essential chemistry knowledge to navigate such challenges as context motivation to learn thinking activity language assessment and maintaining professional expertise If you are a secondary or post secondary teacher of chemistry this book will quickly become a favorite well thumbed resource Professor Hannah Sevia University of Massachusetts Boston

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