

	A	B	C	D	E	F	G	H	I	J	K
1	Region	Item	Sales								
2	East	Apples	2,850								
3	East	Apples	3,200								
4	East	Oranges	2,800								
5	East	Oranges	3,350								
6	East	Oranges	2,500								
7	North	Apples	2,600								
8	North	Apples	2,850								
9	North	Apples	2,550								
10	North	Oranges	2,500								
11	North	Oranges	2,550								
12											
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"Region","Item","Sales"

"East","Apples","2,850"

"East","Apples","3,200"

"East","Oranges","2,800"

"East","Oranges","3,350"

"East","Oranges","2,500"

"North","Apples","2,600"

"North","Apples","2,850"

"North","Apples","2,550"

"North","Oranges","2,500"

"North","Oranges","2,550"

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L Cohen



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web degree of reaction zero reaction stage let us first discuss the special case of zero reaction according to the definition of reaction when $\Lambda = 0$ equation upper reveals that $h_1 = h_2$ and equation lower that $\beta_1 = \beta_2$ fifty percent reaction stage β_2 and the velocity diagram is symmetrical because of symmetrical symmetry it is also

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web the 50 reaction turbine has been used widely and has special significance the velocity diagram of a 50 reaction is symmetrical and for the maximum utilization factor the exit velocity v_2 must be axial figure 2 40 shows a velocity diagram of a 50 reaction turbine and the effect on the utilization factor

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web $p_1 v_1 u_1 v_2$ 2 gz_1 q $p_2 v_2 u_2 v_2$ 2 gz_2 w $p_1 v_1 u_1 h_1$ enthalpy of steam at inlet $p_2 v_2 u_2 h_2$ enthalpy of steam at exit $h_1 v_2$ 2 gz_1 q $h_2 v_2$ 2 gz_2 w generally changes in potential energy are negligible $z_1 z_2$ if no heat is supplied from the surroundings then $q = 0$

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