

In today's activity, you and your group will perform alpha particle decay using a simulation, and then use things you know to answer the questions. Read yourself instructions before doing using the Alpha Decay simulation. Steps/Notes on how to use the simulation is hyperlinked through.

After the simulation opens, click on Alpha Decay. The number that will be investigated is particles (200).

1. How many protons and neutrons does polonium-210 have?

Polonium-210 has 84 protons and 126 neutrons.

2. After polonium-210 decays, what stable nucleus is formed?

The stable nucleus that is formed is lead-206.

3. Draw the stable nucleus design when polonium-210 decays to lead-206. What about the mass number?

Lead-206 has 82 protons and 124 neutrons. Mass number is 206.

3. Pressing what control button to stop the simulation again.

4. Press the simulation button, after the decay counts what do you get? Do you notice anything from the results?

The counts show a decrease from time to time as the simulation goes on.

5. If polonium is a radioactive particle, and is made of two protons and two neutrons, if you know this, given the simulation, could you have predicted the identity of the stable nucleus that is formed?

Yes, because the number of protons and neutrons in the nucleus is conserved, so the stable nucleus must be lead-206.

6. What is the half-life of polonium-210?

Polonium-210 has a half-life of 138.38 days.

Holt Science Beta Decay Sim Answers

A Loxley



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