

**May 2009 TZ1 Question 1**

- (b) For part of her extended essay investigation into the efficiency of the process, a student reacted a pure sample of a vegetable oil (where  $R = C_{17}H_{33}$ ) with methanol. The raw data recorded for the reaction is below.

Mass of oil	=	1013.0 g
Mass of methanol	=	200.0 g
Mass of sodium hydroxide	=	3.5 g
Mass of biodiesel produced	=	811.0 g

The relative molecular mass of the oil used by the student is 885.6. Calculate the amount (in moles) of the oil and the methanol used, and hence the amount (in moles) of excess methanol.

[3]

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**May 2009 TZ2 Question 1**

- (b) A student reacted some salicylic acid with excess ethanoic anhydride. Impure solid aspirin was obtained by filtering the reaction mixture. Pure aspirin was obtained by recrystallization. The following table shows the data recorded by the student.

Mass of salicylic acid used	$3.15 \pm 0.02$ g
Mass of pure aspirin obtained	$2.50 \pm 0.02$ g

- (i) Determine the amount, in mol, of salicylic acid,  $C_6H_4(OH)COOH$ , used. [2]

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- (ii) Calculate the theoretical yield, in g, of aspirin,  $C_6H_4(OCOCH_3)COOH$ . [2]

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**ML Yell**



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Chemistry for the IB Diploma Geoff Neuss, 2001 This concise guide provides the content needed for the Chemistry IB diploma at both Standard and Higher Level It follows the structure of the IB Programme exactly and includes all the options Each topic is presented on its own page for clarity Higher Level material is clearly indicated and there are plenty of practice questions The text is written with an awareness that English might not be the reader's first language *Journal of the Society of Chemical Industry* Society of Chemical Industry (Great Britain), 1925 Includes list of members 1882 1902 proceedings of the annual meetings and various supplements I & EC, 1925 **Mining and Engineering World**, 1909

Papers Presented at the National Technical Conference Society of Plastics Engineers, 1969 *Monthly Catalog of United States Government Publications*, 1962 *British Books*, 1911 *Bibliography of Agriculture*, 1964 **Bulletin**, 1910

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