## Presence of a Catalyne

- OWhat effect does the MnO<sub>2</sub> catalyst have on the rate of evolution of O<sub>2</sub> gas?
- 2. Write a balanced equation for the decomposition of H<sub>2</sub>O<sub>2</sub>-

# E. Concentration of Reactants: Magnesium-Hydrochloric Acid System

| of HCI   | of HCI | mol HCI   | mass of Mg | O <sub>mol Mg</sub>         | mol HCl<br>mol Mg | O'Time (sec) |
|--|--------|-----------|------------|-----------------------------|-------------------|--------------|
| 6.M  | -5m    |           | 0.0301     | DESCRIPTION OF THE PARTY OF | Section 1         |              |
| 4 M  | 5,~    | Section 1 | 0.0188     | Constitution of             |                   |              |
| 3.M  | 5/2    |           | 0.0161     | AMERICAN PROPERTY.          |                   |              |
| 1 M  | 500    |           | 0.0100     |                             | or the second     |              |
| The second secon |        |           |            |                             |                   |              |

- 2. From your graph, predict the time, in seconds, for 5 mg of Mg to react in 5 mL of 2.0 M HCL

| Concentration of | Reactants: Indi | c Acid-Sulfurous | Archel Continue |
|------------------|-----------------|------------------|-----------------|
|                  |                 |                  |                 |

Molar concentration of HIO, \_\_\_\_\_\_ Molar concentration of H<sub>2</sub>SO, \_\_\_\_\_\_

What is the volume (mL) per drop of the HIO, solution? \_\_\_\_\_ Owhat is the volume (mL) per drop of the water? \_\_\_\_\_

| Test<br>Tube | Drops<br>of HIO <sub>3</sub> | mL HIO <sub>3</sub>              | Drops<br>of H <sub>2</sub> O   | mL H <sub>2</sub> O | mL H <sub>2</sub> SO <sub>3</sub> | [HIO <sub>3</sub> ] <sub>6</sub> <sup>6</sup><br>(diluted) | Timse<br>(sec) |
|--------------|------------------------------|----------------------------------|--|---------------------|-----------------------------------|--|----------------|
| 91           |                              | -                                |  |                     | 1.0                               |  |                |
| 2            |                              | THE OWNER OF THE PERSON NAMED IN | PART OF THE PART O | Section 1           | 1.0                               | THE RESERVED   | HOUSE STREET   |
| 3            |                              |                                  |  | 1                   | 1.0                               | P. S. P. S. DOWN CO.                                       |                |
| 4 .          |                              |                                  |  |                     | 1.0                               | THE RESERVE  |                |
| 5            |                              |                                  |  |                     | 1.0                               |  |                |

- L. Opiot [HIO<sub>1</sub>]<sub>0</sub> (y-axis) versus time (x-axis). Instructor's approval of graph:
- 2. How does a change in the molar concentration of HIO, affect the time required for the appearance of the deep-blue I<sub>1</sub>"starch complex?
- 3. Estimate the time, in seconds, for the deep-blue I<sub>1</sub>\*-starch complex to form when 10 drops of 0.01 M HIO<sub>3</sub> are used for the reaction. Assume all other conditions remain constant.

"See footnote 5

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**CL Gary** 

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