

Equilibrium Practice Problems

1. Write the equilibrium expression for each of the following reactions:



$$K = \frac{[\text{NH}_3]^2}{[\text{N}_2] [\text{H}_2]^3}$$



$$K = \frac{[\text{ICl}]^2}{[\text{Cl}_2]}$$



$$K = \frac{[\text{NO}] [\text{O}_2]^{1/2}}{[\text{NO}_2]}$$

2. The dissociation of acetic acid, CH_3COOH , has an equilibrium constant at 25°C of 1.8×10^{-5} . The reaction is



If the equilibrium concentration of CH_3COOH is 0.46 moles in 0.500 L of water and that of CH_3COO^- is 8.1×10^{-3} moles in the same 0.500 L, calculate $[\text{H}^+]$ for the reaction.

$$K = \frac{[\text{CH}_3\text{COO}^-] [\text{H}^+]}{[\text{CH}_3\text{COOH}]} \quad \text{Plug in known values and solve.}$$

$$1.8 \times 10^{-5} = \frac{[8.1 \times 10^{-3} \text{ moles}/0.500 \text{ L}] [\text{H}^+]}{[0.46 \text{ moles}/0.500 \text{ L}]}$$

$$[\text{H}^+] = 1.0 \times 10^{-3} \text{ M}$$

3. Indicate the effect of a catalyst, pressure, temperature and concentration on each of the following on:

	catalyst	pressure	temperature	concentration
a. speed	presence increase	direct relationship	direct relationship	direct relationship
b. pos ⁿ	nothing	based on moles	based on enthalpy of rxn	inverse relationship

Equilibrium Problems Practice

Neil D. Jespersen, Pamela Kerrigan



Equilibrium Problems Practice:

Demand and Supply: 500 Practice Problems Solving for Equilibrium Carl Sutton Mapleton, 2017-05 A collection of 500 practice problems involving finding the equilibrium with supply and demand equations Useful for students and educators Economics at all levels contains math which some students may not be comfortable with This basic text provides some practice at the introductory level of economics where students must solve equilibrium problems presented as a system of equations with supply and demand

Survival Guide to General Chemistry Patrick E. McMahon, Rosemary McMahon, Bohdan Khomtchouk, 2019-02-13 This work evolved over thirty combined years of teaching general chemistry to a variety of student demographics The focus is not to recap or review the theoretical concepts well described in the available texts Instead the topics and descriptions in this book make available specific detailed step by step methods and procedures for solving the major types of problems in general chemistry Explanations instructional process sequences solved examples and completely solved practice problems are greatly expanded containing significantly more detail than can usually be devoted to in a comprehensive text Many chapters also provide alternative viewpoints as an aid to understanding Key Features The authors have included every major topic in the first semester of general chemistry and most major topics from the second semester Each is written in a specific and detailed step by step process for problem solving whether mathematical or conceptual Each topic has greatly expanded examples and solved practice problems containing significantly more detail than found in comprehensive texts Includes a chapter designed to eliminate confusion concerning acid base reactions which often persists through working with acid base equilibrium Many chapters provide alternative viewpoints as an aid to understanding This book addresses a very real need for a large number of incoming freshman in STEM fields **AP**

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Chemical Engineering Principles and Applications Nuggenhalli S. Nandagopal, 2023-05-26 This text provides a clear and concise understanding of the principles and applications of chemical engineering using a rigorous yet easy to follow presentation The coverage is broad and it includes all the relevant concepts such as mass and energy balances mass transfer chemical reaction engineering and many more Elucidation of the principles is further reinforced by examples and practice problems with detailed solutions Firmly grounded in the fundamentals the book maximizes readers capacity to take on new problems and challenges in the field with confidence and conviction Providing a ready reference and review of essential principles and their applications in chemical engineering the book is ideal for undergraduate chemical engineering students as well as practicing engineers preparing for the engineering license exams FE and PE in USA and abroad

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Prevention Practice Catherine Rush Thompson, 2007 The all encompassing Prevention Practice A Physical Therapist's

s Guide to Health Fitness and Wellness successfully details the impact health promotion health protection and the prevention of illness and disability have on increasing the quality and length of a healthy life for individuals across the lifespan Dr Catherine Thompson along with eight contributors all with diverse backgrounds in physical therapy rehabilitation and healthcare present the fundamental health fitness and wellness concepts that are critical for providing preventive care to healthy impaired and at risk populations as outlined in the Guide to Physical Therapist Practice and Healthy People 2010 Prevention Practice includes screening tools for determining risk factors associated with common medical problems as well as resources for implementing prevention practice in clinical and community based settings including planning and marketing a prevention practice Additionally this unique guide offers suggestions for providing appropriate interventions consultation referral education resources and program development Some topics covered include

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distinguish this text from many of the current text offerings It more accurately reflects the curriculum of most Canadian institutions Instructors will find the text sufficiently rigorous while it engages and retains student interest through its accessible language and clear problem solving program without an excess of material that makes most text appear daunting and redundant

Analytical Chemistry Gary D. Christian, Purnendu K. Dasgupta, Kevin A. Schug, 2013-10-07 With the 7th Edition of *Analytical Chemistry* renowned chemists Purnendu K. Dasgupta and Kevin A. Schug both of the University of Texas at Arlington join the author team The new edition focuses on more in depth coverage of the principles and techniques of quantitative analysis and instrumental analysis aka Analytical Chemistry The goal of the text is to provide a foundation of the analytical process tools and computational methods and resources and to illustrate with problems that bring realism to the practice and importance of analytical chemistry It is designed for undergraduate college students majoring in chemistry and in fields related to chemistry

Problems and Problem Solving in Chemistry Education Georgios Tsapralis, 2021-05-17 Problem solving is central to the teaching and learning of chemistry at secondary tertiary and post tertiary levels of education opening to students and professional chemists alike a whole new world for analysing data looking for patterns and making deductions As an important higher order thinking skill problem solving also constitutes a major research field in science education Relevant education research is an ongoing process with recent developments occurring not only in the area of quantitative computational problems but also in qualitative problem solving The following situations are considered some general others with a focus on specific areas of chemistry quantitative problems qualitative reasoning metacognition and resource activation deconstructing the problem solving process an overview of the working memory hypothesis reasoning with the electron pushing formalism scaffolding organic synthesis skills spectroscopy for structural characterization in organic chemistry enzyme kinetics problem solving in the academic chemistry laboratory chemistry problem solving in context team based active learning technology for molecular representations IR spectra simulation and computational quantum chemistry tools The book concludes with methodological and epistemological issues in problem solving research and other perspectives in problem solving in chemistry With a foreword by George Bodner

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