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Rivaliser et coopérer avec ses concurrents

Préface de
Dr Charles J. Fombrun

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oxidation $\text{Co} \rightarrow \text{Co}^{2+} + 2\text{e}^-$ reduction $\text{Fe}^{2+} + 2\text{e}^- \rightarrow \text{Fe}$ 24 $3 \text{ Ag} + \text{Ni} \rightarrow 3 \text{ Ag} + \text{Ni}^{2+}$ oxidation reduction 25 $\text{Cu} + \text{Pb} \rightarrow \text{Cu} + \text{Pb}^{2+}$ oxidation reduction 26 O_2

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