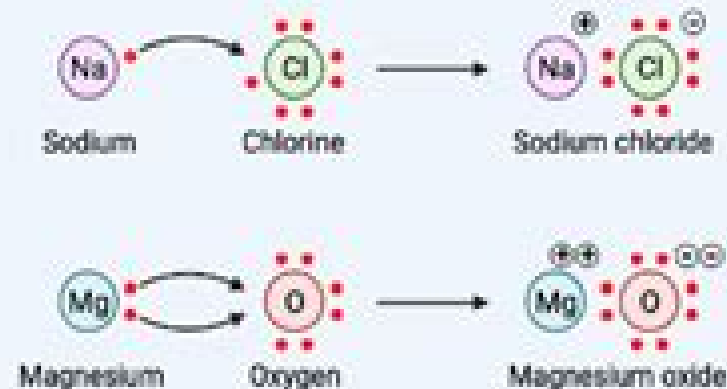


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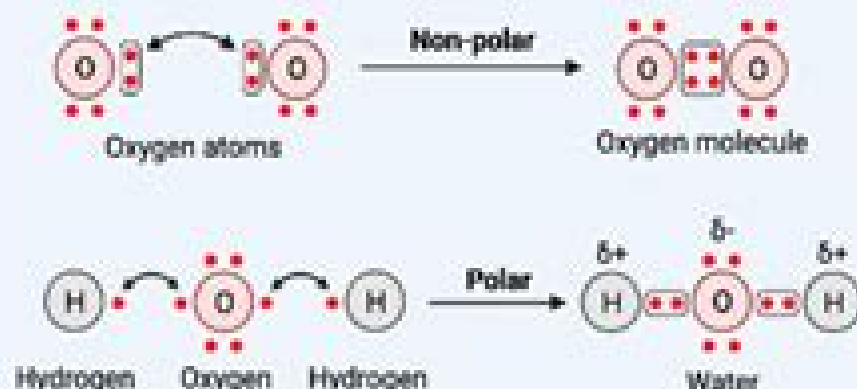
Ionic Bond

Metal atom loses electron(s) to non-metal atom



Covalent Bond

Two non-metal atoms share electrons



Metallic Bond

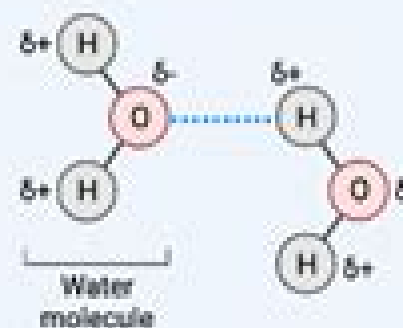
Positive metal ions attract conducting electrons



Copper ions immersed in an electron cloud

Hydrogen Bond

Hydrogen from a molecule attracts an electronegative atom from another molecule electrostatically



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