

ICSE Board Class X Chemistry Board Paper – 2014

Time: 11/2 hrs

Total Marks: 80

General Instructions:

- Answers to this paper must be written on the paper provided separately.
- You will NOT be allowed to write during the first 15 minutes. This time is to be spent in reading the question paper.

The time given at the head of this paper is the time allowed for writing the answers.

This question paper is divided into two sections.

- Section I contains one question with parts (a) to (h); all the eight parts are to be answered.
- Section II contains six questions numbered 2 to 7. You are to answer any four of these
 questions.

The intended marks of questions or for parts of questions are given in brackets [].

SECTION I (40 Marks)

Attempt all questions from this section.

Question 1

(a) Choose the correct answer from the options given below:

[10]

- i. Ionisation potential increases over a period from left to right because the
 - A. Atomic radius increases and nuclear charge increases
 - Atomic radius decreases and nuclear charge decreases
 - C. Atomic radius increases and nuclear charge decreases
 - Atomic radius decreases and nuclear charge increases
- ii. A compound X consists of only molecules. Hence, X will have
 - A. A crystalline hard structure
 - B. A low melting point and low boiling point
 - C. An ionic bond
 - D. A strong force of attraction between its molecules
- iii. When fused lead bromide is electrolysed, we observe
 - A. A silver grey deposit at the anode and a reddish brown deposit at the cathode
 - A silver grey deposit at the cathode and a reddish brown deposit at the anode
 - C. A silver grey deposit at the cathode and reddish brown fumes at the anode
 - Silver grey fumes at the anode and reddish brown fumes at the cathode
- iv. The main ore used for the extraction of iron is
 - A. Haematite
 - B. Calamine
 - C. Bauxite
 - D. Cryolite

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Lingjun Ying

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Table of Contents Guide For Students Appearing For Icse 2014

- 1. Understanding the eBook Guide For Students Appearing For Icse 2014
 - The Rise of Digital Reading Guide For Students Appearing For Icse 2014
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Guide For Students Appearing For Icse 2014
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Guide For Students Appearing For Icse 2014
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Guide For Students Appearing For Icse 2014
 - Personalized Recommendations
 - Guide For Students Appearing For Icse 2014 User Reviews and Ratings
 - Guide For Students Appearing For Icse 2014 and Bestseller Lists
- 5. Accessing Guide For Students Appearing For Icse 2014 Free and Paid eBooks
 - Guide For Students Appearing For Icse 2014 Public Domain eBooks
 - Guide For Students Appearing For Icse 2014 eBook Subscription Services
 - Guide For Students Appearing For Icse 2014 Budget-Friendly Options
- 6. Navigating Guide For Students Appearing For Icse 2014 eBook Formats

- ePub, PDF, MOBI, and More
- Guide For Students Appearing For Icse 2014 Compatibility with Devices
- Guide For Students Appearing For Icse 2014 Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Guide For Students Appearing For Icse 2014
 - Highlighting and Note-Taking Guide For Students Appearing For Icse 2014
 - Interactive Elements Guide For Students Appearing For Icse 2014
- 8. Staying Engaged with Guide For Students Appearing For Icse 2014
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Guide For Students Appearing For Icse 2014
- 9. Balancing eBooks and Physical Books Guide For Students Appearing For Icse 2014
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Guide For Students Appearing For Icse 2014
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Guide For Students Appearing For Icse 2014
 - Setting Reading Goals Guide For Students Appearing For Icse 2014
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Guide For Students Appearing For Icse 2014
 - Fact-Checking eBook Content of Guide For Students Appearing For Icse 2014
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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