

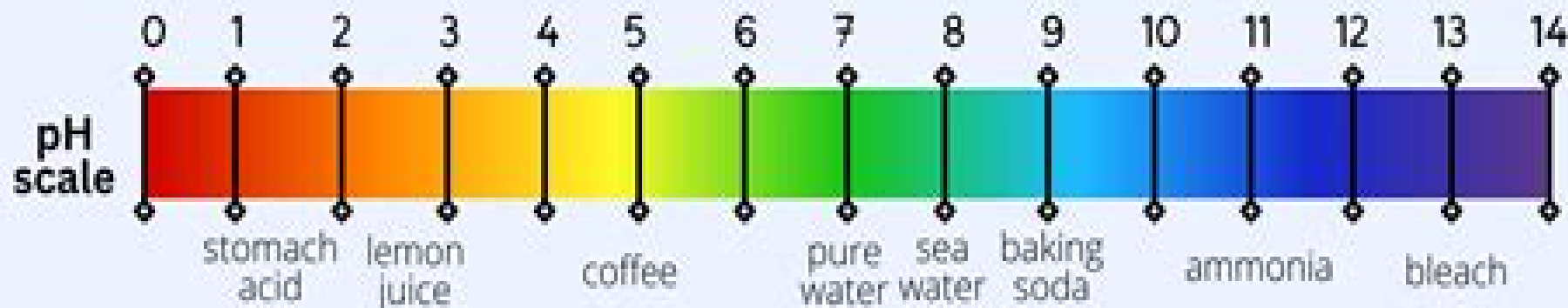
Acid-Base Chemistry

acids

- pH less than 7
- Turn litmus paper red
- Taste sour
- Feel irritating/corrosive
- Include fruit juices, soda, and coffee

bases

- pH greater than 7
- Turn litmus paper blue
- Taste bitter or soapy
- Feel slippery
- Include baking soda, ammonia, and soap



neutralization
reaction

strong acid + **strong base** → salt + water

example: **HCl** + **NaOH** → NaCl + H₂O

Guide Acids Bases And Salts

Michael Strachan

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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Basic Concepts of Chemistry, 9e Study Guide and Solutions Manual Leo J. Malone, Theodore O. Dolter, 2012-01-03 The 9th edition of Malone's Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment. New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections, assessment exercises at the end of each section, and relevant chapter problems at the end of each chapter. A new Math Check allows quick access to the needed basic skill. The first chapter now includes brief introductions to several fundamental chemical concepts, and Chapter Synthesis Problems have been added to the end of each chapter to bring key concepts into one encompassing problem. Every concept in the text is clearly illustrated with one or more step-by-step examples. Making it Real essays have been updated to present timely and engaging real-world applications, emphasizing the relevance of the material they are learning. This edition continues the end-of-chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical aspect of the material discussed in the chapter. *Cambridge IGCSE® Chemistry Practical Teacher's Guide with CD-ROM* Michael Strachan, 2016-06-23 This edition of our successful series to support the Cambridge IGCSE Chemistry syllabus 0620 is fully updated for the revised syllabus from first examination from 2016. The Cambridge IGCSE Chemistry Practical Teacher's Guide complements the Practical Workbook, helping teachers to include more practical work in lessons. Specific support is provided for each of the carefully designed investigations to save teachers time. The Teacher's Guide contains advice about planning investigations, guidance about safety considerations, differentiated learning suggestions to support students who might be struggling, and to stretch the students who are most able, as well as answers to all the questions in the Workbook. The Teacher's Guide also includes a CD-ROM containing model data to be used in instances when an investigation cannot be carried out. *Pocket Guide to Fluid, Electrolyte, and Acid-Base Balance - E-Book* Ursula Heitz, Mima M. Horne, 2004-10-20 Compares different types of parenteral and enteral feedings along with nursing implications. Contains expanded and updated relevant IV therapy and pharmacology. Features updated content throughout with current literature and research findings such as most current blood pressure guidelines from the U.S. Department of Health and Human Services. *The Prospector's Field-book and Guide in the Search for and the Easy Determination of Ores and Other Useful Minerals* Henry Stafford Osborn, 1901 **Laboratory Manual for Principles of General Chemistry** J. A. Beran, Mark Lassiter, 2022-08-16 Laboratory Manual for Principles of General Chemistry 11th Edition covers two semesters of a general chemistry laboratory program. The material focuses on the lab experiences that reinforce the concepts that not all experimental conclusions are the same and depend on identifying an appropriate experimental procedure, selecting the proper apparatus, employing the proper techniques, systematically analyzing and interpreting the data, and minimizing inherent variables. As a result of good data, a scientific and analytical conclusion is made, which may or may not be right, but is certainly consistent with the data.

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