

Cover Page - Service

PRODUCT DESCRIPTION

OPERATING PRINCIPLES

TROUBLE SHOOTING

DISASSEMBLY & ASSEMBLY

ADJUSTMENT

MAINTENANCE

Epson Stylus Pro 7700M/7710M (Cop

APPENDIX

8.1 Block Wiring Diagram

8.2 Panel Menu Map

8.3 Part names used in this manu

8.4 Parts List

8.5 Exploded Diagram

SERVICE MANUAL



Large Format Color Inkjet Printer

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lialh oc ch3 3 h 4 benzaldehyde c6h5cho ch2ohch2oh h step 1 toluene oxidation with kmno4 kmno4 is a strong oxidising
agent which oxidise toluene to benzaldehyde in 1st step

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oxidation of whatever substrate may be present under basic neutral and acidic conditions respectively

$$\text{MnO}_4^- + 4\text{H}^+ + 3\text{e}^- \rightarrow \text{MnO}_2 + 2\text{H}_2\text{O}$$

$$\text{MnO}_4^- + 2\text{H}_2\text{O} + 3\text{e}^- \rightarrow \text{MnO}_2 + 4\text{OH}^-$$

$$\text{MnO}_4^- + 8\text{H}^+ + 5\text{e}^- \rightarrow \text{Mn}^{2+} + 4\text{H}_2\text{O}$$
 basic m n x 7 o x 4 x e x m n x 6 o x 4 x 2 b a s i c

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benzylic hydrogen in toluene and synthesize benzoic acid however i cannot find exact mechanism of reaction in

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notes the position directly adjacent to an aromatic group is called the benzylic position the reaction only works if there is a
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various parameters have been optimized the oxidation of toluene by aq kmno 4 gives benzoic acid scheme 1 kmno 4 in turn is
reduced to mno 2 the reaction did not give any other byproduct

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