

**ঘের ও কীটপতঙ্গের অংশের নির্ধারিত পটীকার প্রস্তুতকরণ**

চলকের ঘনমাত্রা,

$$S = \frac{W}{V} = \frac{20}{36.5} = 1$$

$$M = 0.5479 \text{ M}$$

অতএব, HCl চলকের ঘনমাত্রা

$$= 0.5479 \text{ M}$$

এখানে,

$$\text{মলের ভর, } m = 20 \text{ g}$$

$$\text{HCl এর আপেক্ষিক ভর,}$$

$$M = 36.5 \text{ g/mol}$$

$$\text{চলকের আয়তন, } V = 1 \text{ L}$$

$$\text{চলকের ঘনমাত্রা } S = ?$$

উদাহরণে উল্লিখিত X-পদার্থটিতে 10 mg এর মধ্যে কত পরিমাণ মূল ও Y-পদার্থটিতে 0.5 g 2%  $\left(\frac{W}{V}\right)$  HCl দ্রবণ নিয়ে কাজ করা হয়েছে।

যদিও X-পদার্থটি হলো সেমি-মাইক্রো আনালিটিক্যাল পদার্থ ও Y-পদার্থটি হলো ছোটো আনালিটিক্যাল পদার্থ। উভয় পদার্থের মধ্যে সেমি-মাইক্রো পদার্থটি পরিবেশে ব্যাপক। নিম্নে তা বিশ্লেষণ করা হলো-

সেমি-মাইক্রো পদার্থটি রাসায়নিক গুণগত বিশ্লেষণের একটি পদার্থ। এটি বিশ্লেষণের সময় যত্ন সহকারে রাসায়নিক উপাদান সন্ধানকরণ ও বিশ্লেষণের বিশ্লেষণে ব্যবহৃত হয়। এ পদার্থটিতে খুব কম পরিমাণ (0.05 থেকে 0.2g কঠিন বা 2-4ml তরল দ্রব্য) রাসায়নিক পদার্থ নিয়ে গঠিত করা হয়। ফলে পরিবেশে মূল্যবান উল্লেখযোগ্যভাবে কমে আসে।

এছাড়াও সেমি-মাইক্রো পদার্থটিতে  $\text{H}_2\text{S}$  এর পরিমাণে বৈজ্ঞানিকভাবে ব্যবহার করা হয় ফলে পরিবেশে মূল্য হ্রাস পায় না।

$(\text{CN})_2\text{H}_2 + \text{H}_2\text{O} \rightarrow \text{CH}_3\text{CONH}_2 + \text{H}_2\text{S}$

অতএব, কোনো কোনো ক্ষেত্রে কম স্বীকৃত রাসায়নিক দ্রব্য ব্যবহৃত হয়। ফলে পরিবেশে মূল্যবান সন্ধান কমে আসে। তাই সেমি-মাইক্রো পদার্থটি পরিবেশে ব্যাপক।

**উদাহরণ**



- উদাহরণে মূল্য কী?
- সেমি-মাইক্রো পদার্থটি পরিবেশে ব্যাপক কেন?
- চলকের জন্য প্রয়োজ্য হ্যাঞ্জার প্রকৃতিটি অঙ্কন করা এবং ব্যাখ্যা করা।
- উদাহরণের অংশের মূল্য থেকে উপযুক্ত কাচ পাত্র ব্যবহার করে 1M ঘনমাত্রার মূল্য তৈরি করার প্রক্রিয়া ব্যাখ্যা করা। ৪ নং প্রশ্নের উত্তর

বিশ্লেষণের নিয়ন্ত্রিত কক্ষের উপর রাইজার স্থাপন করলে কক্ষের প্রতি দিকের দৃশ্য যে ভর পাওয়া যায়, তাকে রাইজার দৃশ্য বলা হয়।

সেমি-মাইক্রো পদার্থটি রাসায়নিক গুণগত বিশ্লেষণের একটি পদার্থ। এটি বিশ্লেষণের সময় যত্ন সহকারে রাসায়নিক উপাদান সন্ধানকরণ ও বিশ্লেষণের বিশ্লেষণে ব্যবহৃত হয়। এ পদার্থটিতে খুব কম পরিমাণ (0.05 থেকে 0.2g কঠিন বা 2-4ml তরল দ্রব্য) রাসায়নিক পদার্থ নিয়ে গঠিত করা হয়। ফলে পরিবেশে মূল্যবান উল্লেখযোগ্যভাবে কমে আসে।

অতএব, কোনো কোনো ক্ষেত্রে কম স্বীকৃত রাসায়নিক দ্রব্য ব্যবহৃত হয়। ফলে পরিবেশে মূল্যবান সন্ধান কমে আসে। তাই সেমি-মাইক্রো পদার্থটি পরিবেশে ব্যাপক।

প্রকৃতি	করকারী (Corrosive)
আয়তন	করকারী রাসায়নিক উপাদান কক্ষে যে মাত্রা রাসায়নিক দ্রব্যকে মূল্যবান করে সংশোধিত তরল ও অন্যান্য পদার্থের মাত্রা হ্রাস করে ক্ষতি সাধিত হয়। এতে তরল পুড়ে যায় এবং অন্যান্য পদার্থ পুড়ে যায়।
উপাদান	পাত্র কঠিন এলিমেন্ট (HCl, HNO <sub>3</sub> , H <sub>2</sub> SO <sub>4</sub> ), পাত্র কঠিন (NaOH, KOH), H <sub>2</sub> O <sub>2</sub> , অ্যাসেনিক লিকার প্রভৃতি মূল্য ইত্যাদি।
সংরক্ষণ পাত্র	<ul style="list-style-type: none"> <li>পাত্র এলিমেন্টে পোড়ানো রাখতে হবে।</li> <li>পাত্র কঠিনে, কঠিন পদার্থের পিচ্ছিল পোড়ানো রাখতে হবে।</li> </ul>

উদাহরণের মূল্যটি হলো 10M  $\text{H}_2\text{SO}_4$  মূল্য এবং এর থেকে 1M ঘনমাত্রার মূল্য প্রস্তুত করতে হবে 1ম ও ২ম চলকের ব্যবহার করতে হবে।

1ম চলকের দ্রবণ হলো 100 ml অ্যাসেনিক লিকার ও এর দি হলো পিচ্ছিল। সেমি-মাইক্রো অ্যাসেনিক লিকারের অ্যাসেন 100 ml, তাই অ্যাসেনের 100 ml 1M ঘনমাত্রার  $\text{H}_2\text{SO}_4$  মূল্য তৈরি করতে হবে।

$$V_1S_1 = V_2S_2$$

$$\text{বা, } V_1 = \frac{V_2S_2}{S_1}$$

$$= \left( \frac{100 \times 1}{10} \right) \text{ mL}$$

$$= 10 \text{ mL}$$

অতএব, 10ml  $\text{H}_2\text{SO}_4$  মূল্যের সঙ্গে পানি মিশ্রণ করতে হবে (100-10) mL = 90 mL।

অতএব, পিচ্ছিলের সহায়তায় 10 ml মূল্য ঘনমাত্রার মূল্য নিয়ে তার সাথে অ্যাসেনিক লিকার 90 ml পানি মিশ্রণ করে অ্যাসেনের কঠিন ঘনমাত্রার মূল্য পাওয়া যায়।

**উদাহরণ**



- অ্যাসেনিক লিকার কাকে বলে?
- সামান্য মূল্য K<sub>2</sub> এর মূল্য কখনো অসীম হয় না-ব্যাখ্যা করে।
- A চিত্রের মতো কক্ষে উত্তর করার কৌশল বর্ণনা করে।
- B ও C চিত্রের সাথে সম্পর্কিত রাসায়নিক মূল্যবান ব্যবহারকালে খাখা স্বীকৃত মূল্য উপর সম্পর্কিত বিশ্লেষণ করে। ৪ নং প্রশ্নের উত্তর

অ্যাসেনিক লিকার এক ধরনের রাসায়নিক বা কৃত্রিম দ্রব্য সংরক্ষক যা মূল্য বা উপাদানসমূহকে জারণ প্রতিরোধ করে রাখতে সাহায্য করে।

সামান্য মূল্য K<sub>2</sub> এর মূল্য কখনো অসীম বা শূন্য হতে পারে না। উদাহরণ ব্যাখ্যা করার জন্য নিম্নের উদাহরণ মীমাংসার মাধ্যমে করি-

$$A + B \rightarrow C + D$$

# Hsc Chemistry 1 First Chapter Notes

**K G Saur Books**



## Hsc Chemistry 1 First Chapter Notes:

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