

Gas Law Formulas

$$PV = nRT$$

$$d = \frac{PM_w}{RT} \quad v_{rms} = \sqrt{\frac{3RT}{M_w}}$$

$$\frac{P_1 V_1}{n_1 T_1} = \frac{P_2 V_2}{n_2 T_2}$$

$$KE = \frac{3}{2} RT \quad P_A = X_A P_T$$

$$P_T = P_1 + P_2 + P_3$$

$$\frac{R_2}{R_1} = \sqrt{\frac{M_{w1}}{M_{w2}}} = \frac{t_1}{t_2}$$

Guide Gas Laws

Ian Pickup



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