

Chapter 8: An Introduction to Metabolism

- 8.1 Identify the first two laws of thermodynamics and explain how they relate to biological systems.
- 8.2 Explain what is meant by change in free energy and relate it to the following: spontaneous reactions, capacity to do work, stability, and exergonic and endergonic reactions.
- 8.3 Use examples to illustrate the coupling of ATP hydrolysis to endergonic reactions.
- 8.4 Explain how enzymes speed up metabolic reactions.
- 8.5 Describe how regulation of enzyme activity helps control metabolism.

This chapter on energy transfer and enzyme function is fundamental to your understanding of numerous topics discussed later in the course. Your careful study of this chapter will help you understand biological pathways and the regulation of cellular processes.

Study Tip: As you read this chapter, consider which processes described release energy and which ones require an input of energy.

Concept 8.1 *An organism's metabolism transforms matter and energy*

LO 8.1: *Identify the first two laws of thermodynamics and explain how they relate to biological systems.*

The totality of an organism's chemical reactions is called *metabolism*. As a whole, metabolism manages the material and energy resources of the cell in intersecting pathways.

1. There are two types of reactions in metabolic pathways: *anabolic* and *catabolic*.

	a. Which reactions release energy?
	b. Which reactions consume energy?
	c. Which reactions build up larger molecules?
	d. Which reactions break down molecules?
	e. Which reactions are considered “uphill”?
	f. Which type of reaction is photosynthesis?
	g. Which type of reaction is cellular respiration?
	h. In living systems which reactions require enzymes to catalyze them?

Guide For 12th Biology

Shasha Hu



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Table of Contents Guide For 12th Biology

1. Understanding the eBook Guide For 12th Biology
 - The Rise of Digital Reading Guide For 12th Biology
 - Advantages of eBooks Over Traditional Books
2. Identifying Guide For 12th Biology
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Guide For 12th Biology
 - User-Friendly Interface
4. Exploring eBook Recommendations from Guide For 12th Biology
 - Personalized Recommendations
 - Guide For 12th Biology User Reviews and Ratings
 - Guide For 12th Biology and Bestseller Lists

5. Accessing Guide For 12th Biology Free and Paid eBooks
 - Guide For 12th Biology Public Domain eBooks
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6. Navigating Guide For 12th Biology eBook Formats
 - ePub, PDF, MOBI, and More
 - Guide For 12th Biology Compatibility with Devices
 - Guide For 12th Biology Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Guide For 12th Biology
 - Highlighting and Note-Taking Guide For 12th Biology
 - Interactive Elements Guide For 12th Biology
8. Staying Engaged with Guide For 12th Biology
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Guide For 12th Biology
9. Balancing eBooks and Physical Books Guide For 12th Biology
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Guide For 12th Biology
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Guide For 12th Biology
 - Setting Reading Goals Guide For 12th Biology
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Guide For 12th Biology
 - Fact-Checking eBook Content of Guide For 12th Biology
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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