

# Chapter Test Chapter 10: Matter and Energy

## Matter and Energy

Write each answer in the space provided on the paper at the back of this notebook. Show all work for problems that require calculations.

1. ☐ 1. The equation below defines a process that is endothermic.
  - a. melting
  - b. boiling
  - c. condensing
  - d. sublimating
2. ☐ 2. The equation below is exothermic.
  - a. the condensation of steam into liquid water
  - b. the condensation of steam into liquid water
  - c. the condensation of steam into liquid water
  - d. the condensation of steam into liquid water
3. ☐ 3. The equation below shows the reaction between a solid and a liquid. Which process is occurring?
  - a. condensation
  - b. evaporation
  - c. sublimation
  - d. deposition
4. ☐ 4. Which of the following processes is an endothermic process?
  - a. melting
  - b. condensation
  - c. sublimation
  - d. deposition
5. ☐ 5. Which of the following is not a phase change?
  - a. melting
  - b. boiling
  - c. condensation
  - d. sublimation
6. ☐ 6. The equation below shows the reaction between a solid and a liquid. Which process is occurring?
  - a. condensation
  - b. evaporation
  - c. sublimation
  - d. deposition
7. ☐ 7. The equation below shows the reaction between a solid and a liquid. Which process is occurring?
  - a. condensation
  - b. evaporation
  - c. sublimation
  - d. deposition
8. ☐ 8. The equation below shows the reaction between a solid and a liquid. Which process is occurring?
  - a. condensation
  - b. evaporation
  - c. sublimation
  - d. deposition
9. ☐ 9. The equation below shows the reaction between a solid and a liquid. Which process is occurring?
  - a. condensation
  - b. evaporation
  - c. sublimation
  - d. deposition

# Holt Chemistry Test Solutions Answers

**SJ Ball**



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## **Holt Chemistry Test Solutions Answers** Book Review: Unveiling the Power of Words

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