# H:C:::C:H or H:C≡C:H

#### Example 5

Draw the Lewis diagram for each molecule.

- II. 280
- 2. CH<sub>2</sub>O (The carbon atom is the central atom.)

See harrico m.

The bond between the two nitrogen atoms is a triple bond. The Lewis diagram for N<sub>2</sub> is as follows:

NIINI Or NIINI

In CH<sub>2</sub>O, the central atom is surrounded by two different types of atoms. The Lewis diagram that fills each atom's valence electron shell is as follows:



#### Note

One application of CH<sub>2</sub>O, also called formaldehyde, is the preservation of biological specimens. Aqueous solutions of CH<sub>2</sub>O are called formalin and have a sharp, characteristic (pungent) order.

#### Skill-Building Exercise

Draw the Lewis diagram for each molecule.

- $\mathbf{L} = \mathbf{O}_{2}$
- 2 C.H.

#### Concept Review Exercise

1. What is one clue that a molecule has a multiple bond?

#### Answer

 If single bonds between all atoms do not give all atoms (except hydrogen) an octet, multiple covalent bonds may be present.

#### Kiey Tallogarwa y

# **Holt Biology Skills Answers**

J Dewey

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