

## Gas Laws Worksheet

$$\text{atm} = 760.0 \text{ mm Hg} = 101.3 \text{ kPa} = 760.0 \text{ torr}$$

### Boyle's Law Problems: $P_1 V_1 = P_2 V_2$

1. If 22.5 L of nitrogen at 748 mm Hg are compressed to 725 mm Hg at constant temperature. What is the new volume?

$$(748 \text{ mm Hg})(22.5 \text{ L}) = (725 \text{ mm Hg}) V_2$$

$$V_2 = \frac{(748 \text{ mm Hg})(22.5 \text{ L})}{(725 \text{ mm Hg})}$$

$$V_2 = 23.2 \text{ L}$$

2. A gas with a volume of 4.0 L at a pressure of 205 kPa is allowed to expand to a volume of 12.0 L. What is the pressure in the container if the temperature remains constant?

$$(4.0 \text{ L})(205 \text{ kPa}) = (12.0 \text{ L}) P_2$$

$$P_2 = \frac{(4.0 \text{ L})(205 \text{ kPa})}{12.0 \text{ L}}$$

$$P_2 = 68.3 \text{ kPa}$$

3. What pressure is required to compress 196.0 liters of air at 1.00 atmosphere into a cylinder whose volume is 26.0 liters?

$$(196.0 \text{ L})(1.00 \text{ atm}) = (26.0 \text{ L}) P_2$$

$$P_2 = \frac{(196.0 \text{ L})(1.00 \text{ atm})}{26.0 \text{ L}}$$

$$P_2 = 7.54 \text{ atm}$$

4. A 40.0 L tank of ammonia has a pressure of 12.7 kPa. Calculate the volume of the ammonia if its pressure is changed to 8.4 kPa while its temperature remains constant.

$$(40.0 \text{ L})(12.7 \text{ kPa}) = (8.4 \text{ kPa}) V_2$$

$$V_2 = \frac{(40.0 \text{ L})(12.7 \text{ kPa})}{8.4 \text{ kPa}}$$

$$V_2 = 60.5 \text{ L}$$

# Exploring Gas Laws Worksheet Answers

**K Payea**



## **Exploring Gas Laws Worksheet Answers:**

## Whispering the Strategies of Language: An Psychological Quest through **Exploring Gas Laws Worksheet Answers**

In a digitally-driven earth wherever displays reign great and quick connection drowns out the subtleties of language, the profound strategies and emotional subtleties hidden within words frequently go unheard. However, located within the pages of **Exploring Gas Laws Worksheet Answers** a fascinating fictional value blinking with organic thoughts, lies a fantastic quest waiting to be undertaken. Composed by an experienced wordsmith, this wonderful opus attracts readers on an introspective journey, lightly unraveling the veiled truths and profound affect resonating within ab muscles fabric of each word. Within the psychological depths of the moving evaluation, we can embark upon a honest exploration of the book is key themes, dissect its fascinating writing model, and succumb to the powerful resonance it evokes serious within the recesses of readers hearts.

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