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# Explorer® Digital Set-Top User's Guide

Models 4240HDC™ and 4250HDC™  
with Multi-Stream CableCARD™ Interface



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# Explorer 4250hdc Manual

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web a write a balanced equation for the reaction  $\text{Al} + \text{O}_2 \rightarrow \text{Al}_2\text{O}_3$  4 Al 3  $\text{O}_2 \rightarrow 2\ \text{Al}_2\text{O}_3$  b determine the theoretical yield of  $\text{Al}_2\text{O}_3$   $160\ 0\ \text{g}\ \text{Al} \times 1\ \text{mol}\ \text{Al} / 27\ 0\ \text{g}\ \text{mol}^{-1} \times 2\ \text{mol}\ \text{Al}_2\text{O}_3 / 4\ \text{mol}\ \text{Al} \times 101\ 96\ \text{g}\ \text{mol}^{-1} = 302\ 3\ \text{g}\ \text{Al}_2\text{O}_3$  c determine the percent yield  $260\ 0\ \text{g} / 302\ 3\ \text{g} \times 100 = 86\ 01\%$  theoretical 302 3 g

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